

The hydrogenation of ricinoleic acid with hydrazine hydrate. J. Vofick. Chem. Listy 38, 57-82 (1944).—Ricinoleic acid having an I no. 85.07, mixed with hydrazine hydrate, formed a transparent soap which became white, opaque and insol. with time, the reaction proceeding more freely than with oleic acid. After standing 20 days, the solid  $\alpha$ -hydroxystearic acid (I) sepd. from the unaffected ricinoleic acids with much difficulty, but by satg. with dry HCl a boiling KOH soln. of the mixt., Et hydroxystearate formed, which, crystd. from EtOH and then from  $C_6H_6$ , m. 82.6-83.0°. From a K map of the ester decomptd. by EtOH, a pure form of I resulted which was readily sol. in ether, crystd. from EtOH in long, capillary threads m. 81-83°, showed an acid no. 187.80, Ac no. 193.7, I no. 0.26-0.28% Ag and an analysis corresponding to  $(C_{18}H_{34}O_3)_2$ . The Mg ester crystd. from  $C_6H_6$  as silvery, rhombic plates m. 87-88°. The lit ester crystd. as plates or needles m. 82.5-3.0°. A Pr ester crystd. as plates or  $C_6H_6$  which formed V-shaped clusters m. 80-81°. The isopropyl ester crystd. in scales from  $C_6H_6$  but in needles if rapid evapn. occurred; it m. 47-7.5°. An isobutyl ester crystd. in needle form m. 40°. The lauroyl ester m. 35° and formed needles in  $C_6H_6$ . I, prep'd. by hydrogenating ricinoleic acid with H in the presence of Pt black, was identical with the prepa. made with hydrazine hydrate.

**Frank March**

**APPROVED FOR RELEASE: 03/14/2001**

CIA-RDP86-00513R001860810014-2"

**Hydrants of 12-hydroxystearic acid** and some of its derivatives. J. Vurick. Collection Czechoslov. Chem. Communications 5, 400-417 (1933).—12-Hydroxystearic acid,  $\text{C}_8\text{H}_{14}\text{CH}(\text{OH})(\text{CH}_2)_6\text{CONHNH}_2$ , I, treated with  $\text{NaBH}_4\text{H}_2\text{O}$ , gave II, m. 113.8-10.5°; warmed with a slight excess of  $\text{Ac}_2\text{O}$ , it gave  $\text{C}_8\text{H}_{14}\text{CH}(\text{OH})(\text{CH}_2)_6\text{CONHNHAc}$ , m. 144.5°; when refluxed with  $\text{Ac}_2\text{O}$  for 15 min. it gave  $\text{C}_8\text{H}_{14}\text{CH}(\text{OAc})(\text{CH}_2)_6\text{CONHNHAc}$ , I (2 g.) added slowly to I (12 g.) in  $\text{RIOH}$  and then treated with a g. more of I at once gave III, m. 181-4°. White

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cd

ASH-SEA METALLURGICAL LITERATURE CLASSIFICATION

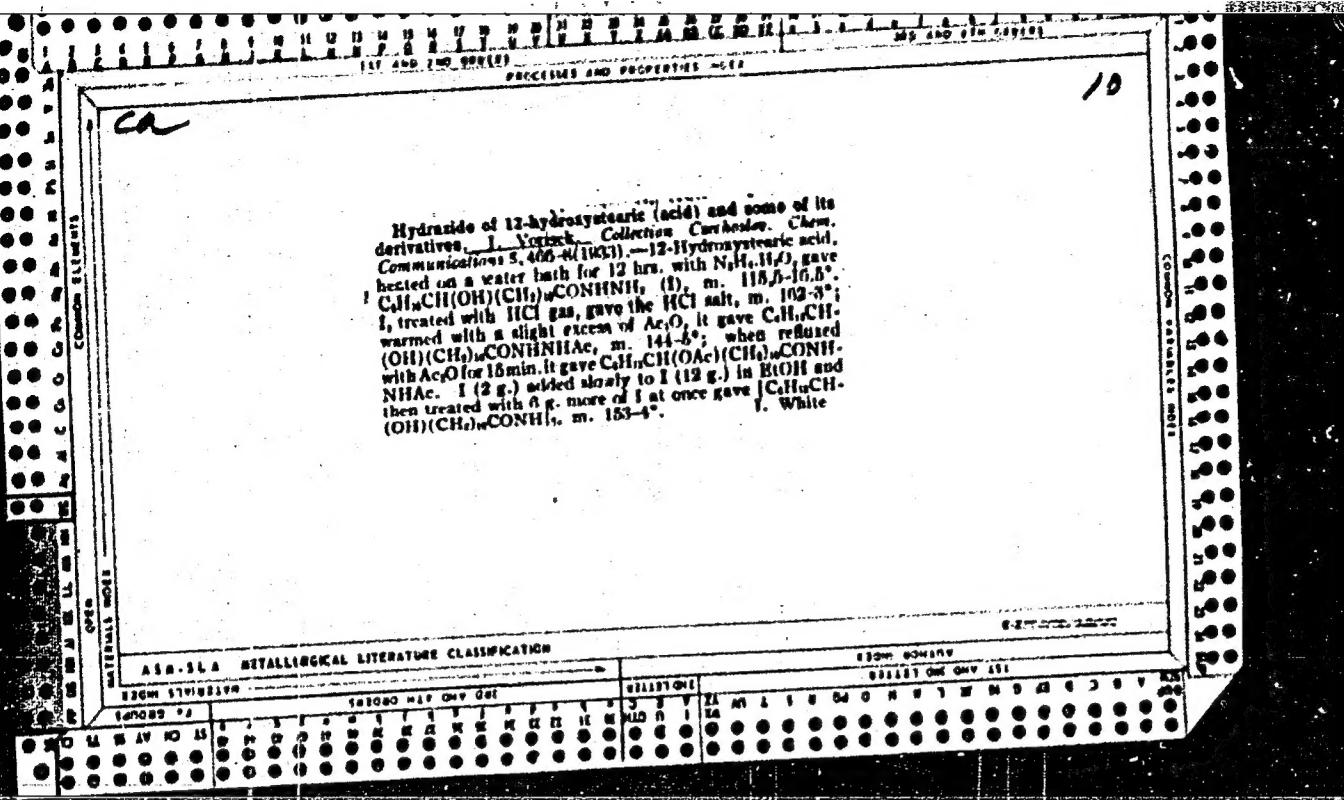
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<p><b>Hydrazide of 12-hydroxystearic (acid) and some of its derivatives.</b> J. Vojick. <i>Collection Czechoslov. Chem. Communications</i> 15, 407 (1950). -12-Hydroxystearic acid, heated on a water bath for 12 hrs. with <math>\text{HgI}_2\text{H}_2\text{O}</math>, gave <math>\text{C}_{18}\text{H}_{34}\text{NO}_2(\text{CH}_2)_6\text{CONHNH}_2</math> (I), m. 112.5-114.5°. I, treated with <math>\text{HCl}</math> gas, gave the <math>\text{HCl}</math> salt, m. 102.3-5°; warmed with a slight excess of <i>AcOH</i>, it gave <math>\text{C}_{18}\text{H}_{32}\text{O}_2(\text{CH}_2)_6\text{CONHNHAc}</math>, m. 141.5°; when refluxed with <i>AcOH</i> for 16 min., it gave <math>\text{C}_{18}\text{H}_{30}\text{O}_2(\text{CH}_2)_6\text{CONHNHAc}</math>. I (2 g.) added slowly to I (12 g.) in <i>EtOH</i> and then treated with 0.5 mole of <i>Li</i> at 0°C gave <math>[\text{C}_{18}\text{H}_{32}\text{O}_2(\text{CH}_2)_6\text{CONH}]_2</math>, m. 133.4°. J. White.</p>																																																																																																													
<p><b>ASU-LLA METALLURGICAL LITERATURE CLASSIFICATION</b></p> <table border="1"> <thead> <tr> <th>EDITION NUMBER</th> <th>SEARCH KEY</th> <th>EDITION NUMBER</th> <th>SEARCH KEY</th> </tr> </thead> <tbody> <tr> <td>SEARCH #1</td> <td>SEARCH #2</td> <td>SEARCH #3</td> <td>SEARCH #4</td> </tr> <tr> <td>SEARCH #5</td> <td>SEARCH #6</td> <td>SEARCH #7</td> <td>SEARCH #8</td> </tr> <tr> <td>SEARCH #9</td> <td>SEARCH #10</td> <td>SEARCH #11</td> <td>SEARCH #12</td> </tr> <tr> <td>SEARCH #13</td> <td>SEARCH #14</td> <td>SEARCH #15</td> <td>SEARCH #16</td> </tr> <tr> <td>SEARCH #17</td> <td>SEARCH #18</td> <td>SEARCH #19</td> <td>SEARCH #20</td> </tr> <tr> <td>SEARCH #21</td> <td>SEARCH #22</td> <td>SEARCH #23</td> <td>SEARCH #24</td> </tr> <tr> <td>SEARCH #25</td> <td>SEARCH #26</td> <td>SEARCH #27</td> <td>SEARCH #28</td> </tr> <tr> <td>SEARCH #29</td> <td>SEARCH #30</td> <td>SEARCH #31</td> <td>SEARCH #32</td> </tr> <tr> <td>SEARCH #33</td> <td>SEARCH #34</td> <td>SEARCH #35</td> <td>SEARCH #36</td> </tr> <tr> <td>SEARCH #37</td> <td>SEARCH #38</td> <td>SEARCH #39</td> <td>SEARCH #40</td> </tr> <tr> <td>SEARCH #41</td> <td>SEARCH #42</td> <td>SEARCH #43</td> <td>SEARCH #44</td> </tr> <tr> <td>SEARCH #45</td> <td>SEARCH #46</td> <td>SEARCH #47</td> <td>SEARCH #48</td> </tr> <tr> <td>SEARCH #49</td> <td>SEARCH #50</td> <td>SEARCH #51</td> <td>SEARCH #52</td> </tr> <tr> <td>SEARCH #53</td> <td>SEARCH #54</td> <td>SEARCH #55</td> <td>SEARCH #56</td> </tr> <tr> <td>SEARCH #57</td> <td>SEARCH #58</td> <td>SEARCH #59</td> <td>SEARCH #60</td> </tr> <tr> <td>SEARCH #61</td> <td>SEARCH #62</td> <td>SEARCH #63</td> <td>SEARCH #64</td> </tr> <tr> <td>SEARCH #65</td> <td>SEARCH #66</td> <td>SEARCH #67</td> <td>SEARCH #68</td> </tr> <tr> <td>SEARCH #69</td> <td>SEARCH #70</td> <td>SEARCH #71</td> <td>SEARCH #72</td> </tr> <tr> <td>SEARCH #73</td> <td>SEARCH #74</td> <td>SEARCH #75</td> <td>SEARCH #76</td> </tr> <tr> <td>SEARCH #77</td> <td>SEARCH #78</td> <td>SEARCH #79</td> <td>SEARCH #80</td> </tr> <tr> <td>SEARCH #81</td> <td>SEARCH #82</td> <td>SEARCH #83</td> <td>SEARCH #84</td> </tr> <tr> <td>SEARCH #85</td> <td>SEARCH #86</td> <td>SEARCH #87</td> <td>SEARCH #88</td> </tr> <tr> <td>SEARCH #89</td> <td>SEARCH #90</td> <td>SEARCH #91</td> <td>SEARCH #92</td> </tr> <tr> <td>SEARCH #93</td> <td>SEARCH #94</td> <td>SEARCH #95</td> <td>SEARCH #96</td> </tr> <tr> <td>SEARCH #97</td> <td>SEARCH #98</td> <td>SEARCH #99</td> <td>SEARCH #100</td> </tr> </tbody> </table>						EDITION NUMBER	SEARCH KEY	EDITION NUMBER	SEARCH KEY	SEARCH #1	SEARCH #2	SEARCH #3	SEARCH #4	SEARCH #5	SEARCH #6	SEARCH #7	SEARCH #8	SEARCH #9	SEARCH #10	SEARCH #11	SEARCH #12	SEARCH #13	SEARCH #14	SEARCH #15	SEARCH #16	SEARCH #17	SEARCH #18	SEARCH #19	SEARCH #20	SEARCH #21	SEARCH #22	SEARCH #23	SEARCH #24	SEARCH #25	SEARCH #26	SEARCH #27	SEARCH #28	SEARCH #29	SEARCH #30	SEARCH #31	SEARCH #32	SEARCH #33	SEARCH #34	SEARCH #35	SEARCH #36	SEARCH #37	SEARCH #38	SEARCH #39	SEARCH #40	SEARCH #41	SEARCH #42	SEARCH #43	SEARCH #44	SEARCH #45	SEARCH #46	SEARCH #47	SEARCH #48	SEARCH #49	SEARCH #50	SEARCH #51	SEARCH #52	SEARCH #53	SEARCH #54	SEARCH #55	SEARCH #56	SEARCH #57	SEARCH #58	SEARCH #59	SEARCH #60	SEARCH #61	SEARCH #62	SEARCH #63	SEARCH #64	SEARCH #65	SEARCH #66	SEARCH #67	SEARCH #68	SEARCH #69	SEARCH #70	SEARCH #71	SEARCH #72	SEARCH #73	SEARCH #74	SEARCH #75	SEARCH #76	SEARCH #77	SEARCH #78	SEARCH #79	SEARCH #80	SEARCH #81	SEARCH #82	SEARCH #83	SEARCH #84	SEARCH #85	SEARCH #86	SEARCH #87	SEARCH #88	SEARCH #89	SEARCH #90	SEARCH #91	SEARCH #92	SEARCH #93	SEARCH #94	SEARCH #95	SEARCH #96	SEARCH #97	SEARCH #98	SEARCH #99	SEARCH #100
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Hydrates of 12-hydroxytaric acid and some of its derivatives. J. Vierck. Collection Comptes Rendus. Chem. Communications 5, 400-4 (1903). - 12-Hydroxytaric acid, heated on a water bath for 12 hrs. with  $\text{NH}_3 \cdot \text{H}_2\text{O}$ , gave  $\text{C}_{10}\text{H}_8\text{CH}(\text{OH})(\text{CH}_3)_2\text{CONHNH}_2$ , (I), m. 118°-16.6°. I, treated with HCl gas, gave the HCl salt, m. 102.3°-5°, warmed with a slight excess of Ac<sub>2</sub>O, it gave  $\text{C}_{10}\text{H}_8\text{CH}(\text{OH})(\text{CH}_3)_2\text{CONHNHAc}$ , m. 144-6°; when refluxed with Ac<sub>2</sub>O for 18 min. it gave  $\text{C}_{10}\text{H}_8\text{CH}(\text{OAc})(\text{CH}_3)_2\text{CONHNHAc}$ . I (2 g.) added slowly to I (12 g.) in EtOH and then treated with a g. more of I at once gave  $\text{J}[\text{C}_{10}\text{H}_8\text{CH}(\text{OH})(\text{CH}_3)_2\text{CONH}]_2$ , m. 183-4°. J. White.



CA

The separation of uranium from cobalt and nickel with the aid of laetic  $\beta$ -oxime. V. Horváth and J. Vodálka, *Chem. Listy* 44, 75 (1950).—To 20-80 cc. of soln. contg. 0.012-0.1 g. of U, 0.015-0.3 g. of Co<sup>2+</sup>, and 0.03-0.3 g. of Ni<sup>2+</sup>, add 10-15 cc. of a buffer (a 10% NaCONa soln. made acid to phosphomolybdate), 5-15 cc. of a 2% NHCLCNS soln., 2.5-15 cc. of a 2% Na K tartrate soln., and 8-20 cc. of a 1% laetic soln. in 50% EtOH. After the yellow ppt. of the uranylatoxime has settled for 15 min., filter and wash the ppt. with about 100 cc. of a 0.25% laetic soln. Ignite the ppt. and weigh the residual UO<sub>3</sub>. The sepn. is based upon the fact that at room temp., the UO<sub>3</sub><sup>++</sup> forms a ppt. with the laetic whereas Co and Ni do not. With slight modifications the method can be used for sepn. U from Mn, Zn, Mg, Ca, Sr and Ba or from their mixts. The errors ranged from +0.0002 to +0.003 g. F. M.

## **1.1.4 METALLURGICAL LITERATURE CLASSIFICATION**

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CIA-RDP86-00513R001860810014-2"

*Ca*

10

The *N*-aminotriazoles of some higher aliphatic acids.  
 J. Am. Chem. Soc., Collection Cechetov, Chem. Communications, No. 6, 99-100 (1934). — *Bis(1-hydroxyheptadecyl)-N-aminotriazole* was prep'd. by reducing  $C_{17}H_{35}CH(OH)(CH_2)_7CO_2H$  and  $H_2NNH_2$  for 50 hrs. on a water bath. The solid product, washed with water and crystd. from alc., m. 139.5-40.6°. The Ac. deriv. has an Ac. no. of 215.9 (theory 216.3). The acetylester, m. 144-45°, m.p. 109.4 (theory 109.3). The acetylester derivative can be obtained, since at the temp. necessary for hydrolysis of the  $NH_2$  group the  $OH$  groups' dissociate. The HCl salt, m. 105-6°, the  $H_2SO_4$  salt 100.5-7.5°. Normal  $C_{17}H_{35}CONHNH_2$  and  $H_2NNH_2$ .

refined 70 hrs., yielded *bis(heptadecyl)-N-aminotriazole*,  
 m. 135.5-0.0°. The HCl salt loses HCl on heating. The  
 Ac. deriv. m. 87-8°. — W. A. Moore

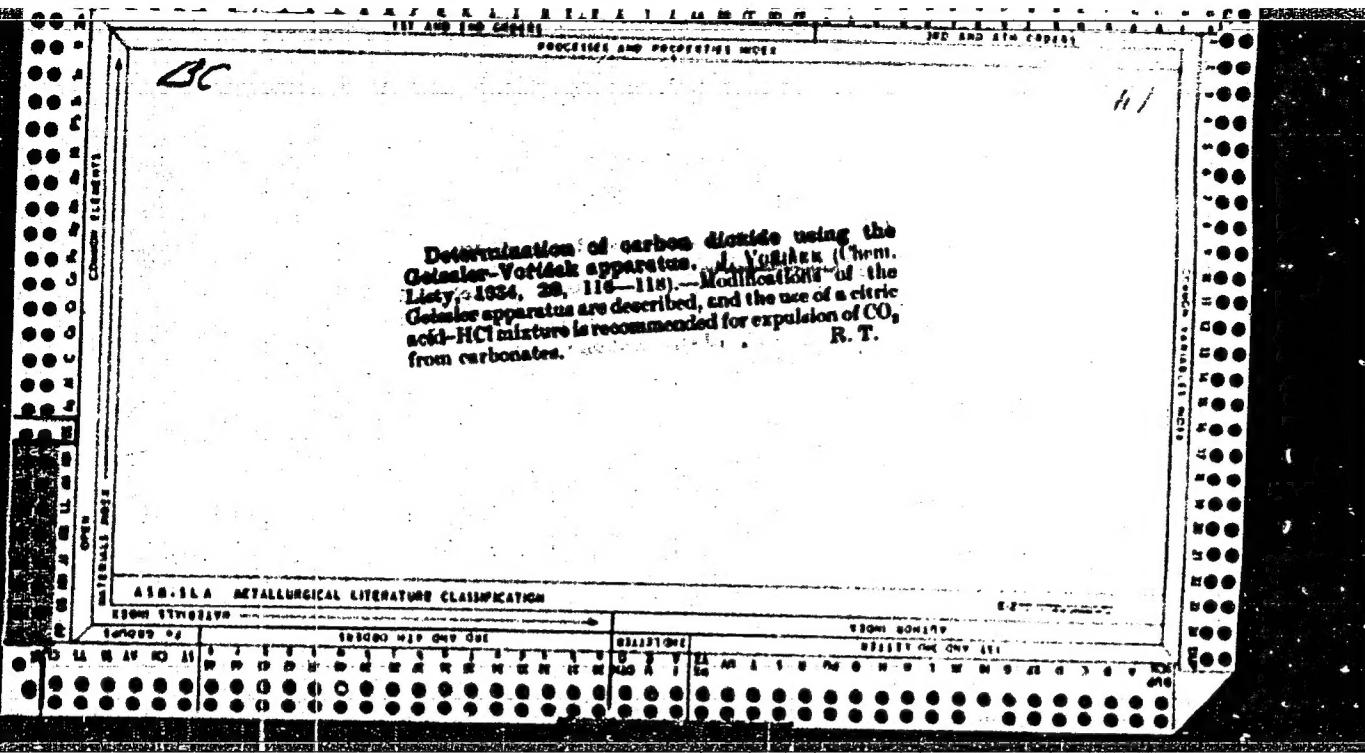
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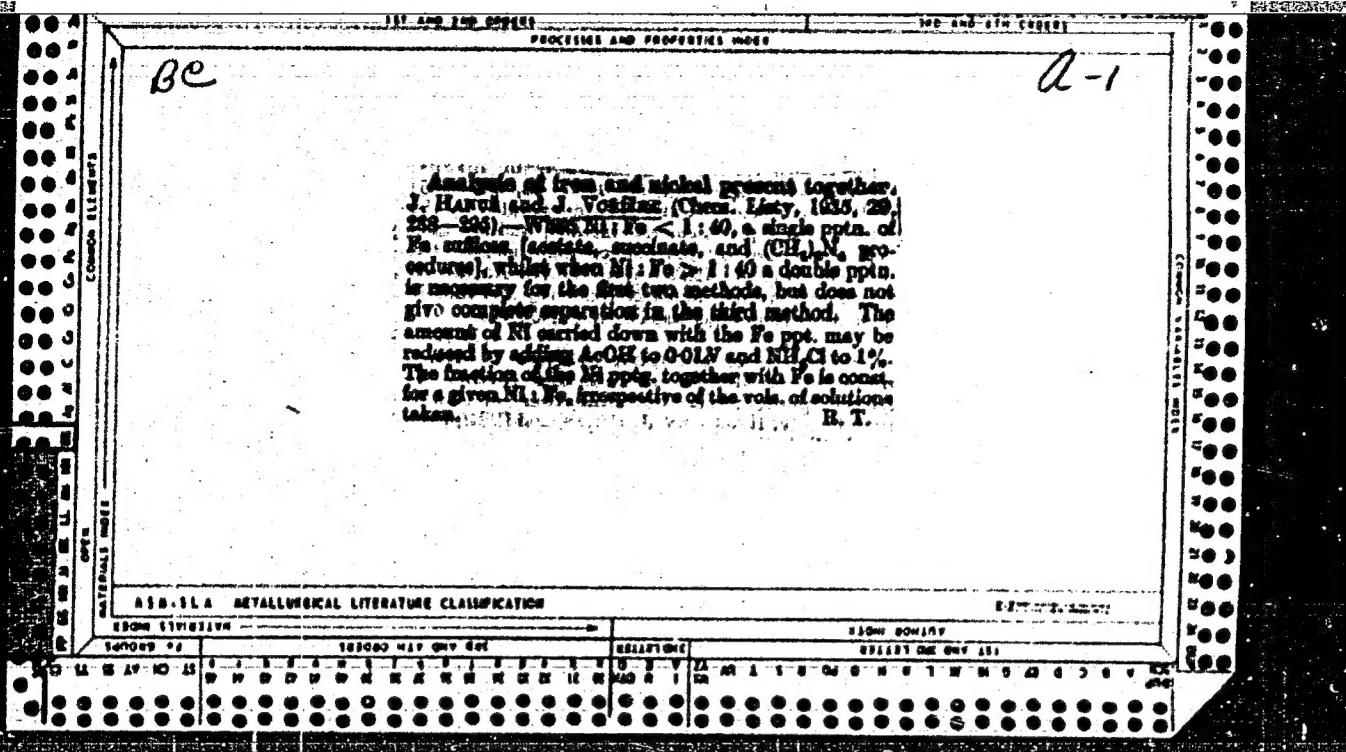
ECONOMICS

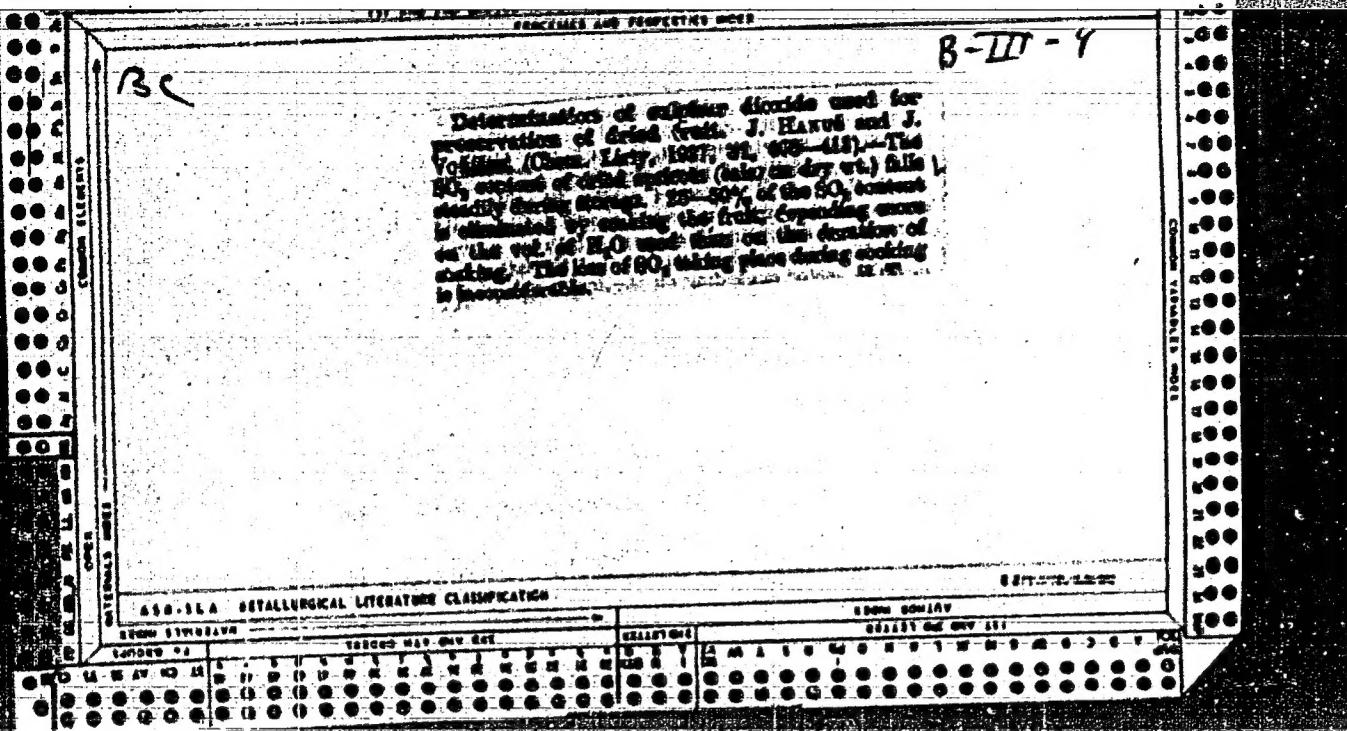
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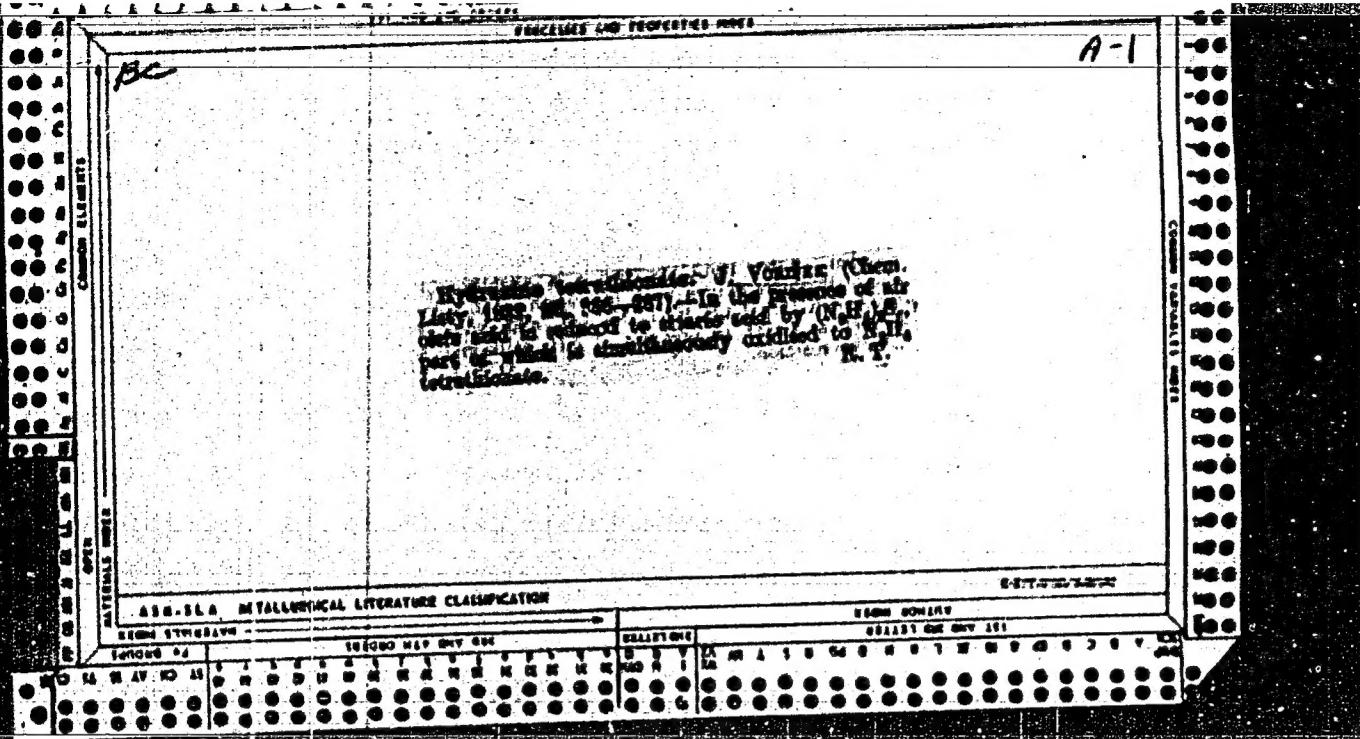
ECONOMICS

TECHNIQUE









*(3c)*

Determination of cobalt-chromium compounds by polarimetry. J. Metallurg. Chem. Ind., 1930, 24, 520-521. (The following method is described for the determination of Cobalt in bearing metal: 0.5 g. of alloy is dissolved in 10 c.c. of hot concentrated sulphuric acid; and 150 c.c. of water; 10 c.c. of 3% potassium hydrogen sulphate solution, and 10 c.c. of concentrated hydrochloric acid are added. The solution is boiled for 5 min. to remove sulphur dioxide, cooled, and 10 c.c. of 10% manganese sulphate (tetrahydrate) are added; after which the solution is titrated with 0.1N potassium permanganate.)

## A88-344 METALLURGICAL LITERATURE CLASSIFICATION

SEARCHED *1960 SEP 1961*SERIALIZED *1960 SEP 1961*INDEXED *1960 SEP 1961*FILED *1960 SEP 1961*SEARCHED *1960 SEP 1961*SERIALIZED *1960 SEP 1961*INDEXED *1960 SEP 1961*FILED *1960 SEP 1961*

1,2-dihydro-3-hydroxy-4-oxo-5-phenyl-1,2-dihydro-*p*-phenylenehydrazide and derivative. J. Voelker (Coll. Czech. Chem. Comm., 1933, 8, 466-470)---Diphenylhydrazide, m.p. 116.5-116.8°; Ammonium salt, m.p. 162-163°;  $\text{NaOAc}$ , m.p. 144-145° (warm  $\text{Ac}_2\text{O}$  and  $\text{EtOH}$ ), and  $\text{UNa}^+$  derivative ( $\text{AcO}^-$  at b.p.), is formed from the acid and  $\text{NH}_2\text{H}_2\text{O}$  at 100°, and with  $\text{Li}$  in  $\text{EtOH}$  gives 2,4-dihydro-3-hydroxy-5-phenyl-1,2-dihydro-*p*-phenylenehydrazine, m.p. 153-154°. H. A. P.

2-3

APPROVED FOR RELEASE: 03/14/2001 CIA-RDP86-00513R001860810014-2"

The determination of carbon dioxide in carbonates with the Geelze-Volhard apparatus. — I. Method. — II. Application of the method to the determination of carbon dioxide in uranite and an alkalimeter is described and discussed. V. M.

4

7

## **APPENDIX B METALLURGICAL LITERATURE CLASSIFICATION**

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CIA-RDP86-00513R001860810014-2"

11

Determination of Copper with isoNitroso-3-phenylpyrazolone. V.  
 Hovorka and J. Vorisek (Chem. Listy, 1942, 36, 73-78; Chem. Zentr., 1942  
 113, (11), 573; C. Abs., 1943, 37, 432).—After exhaustive study of the  
 properties of the insoluble, microcrystalline, brownish-green salt of  
 the composition  $Cu(C_6H_5N_3O_2)_2$ , 2 methods of carrying out the analysis  
 are recommended, by which fairly exact values can be obtained. (1) Tartrate method.

either of which satisfactory values can be obtained. (1) *Tartrate method*. Neutralize with  $\text{NH}_3$ , the acid solution containing about 0.1 gms.  $\text{Cu}(\text{AsO}_4)_2$ , chloride, or nitrate, and then add 5 c.c. of 0.05*N*- $\text{H}_3\text{PO}_4$ , and 2.5-3 gms. of  $\text{NH}_4$  tartrate. Dilute with water to 100-150 cc., heat to 50°C., and add 25 (0 c.c. of a 1% solution of the reagent in methyl alcohol or in 50% hot ethyl alcohol. The supernatant liquid above the brownish green precipitate should be yellowish-brown. After 12-24 hrs., filter through paper, wash with cold  $\text{NH}_3$  tartrate solution, add oxalic acid to the moist precipitate, ignite, and weigh as  $\text{CuO}$ . (2) *Acetate method*. Proceed as above, but after the neutralization with  $\text{NH}_3$  and slight acidification, add 10-15 c.c. of 10%  $\text{Na}\text{ acetate}$  solution which has been made neutral to phenolphthalein.

~~ADD 81A BIBLIOGRAPHICAL LITERATURE CLASSIFICATION~~

**APPROVED FOR RELEASE: 03/14/2001**

CIA-RDP86-00513R001860810014-2"

BC

R-3

N-Aminotriacetic acid. J. Volmer (Chit. Comit. Chem. Oscany) (324, 6, 62) described heating of the higher aliphatic acids (H<sub>3</sub>C)<sub>n</sub>COOH, in addition to the hydrides, under vacuum at 150°C. Below appear derivatives of Fmoc-4-aminobutyric acid: m.p. 129.5—140.0°; 125 (neutral) and 126 (acidic); hydrochloride, m.p. 130.5—131.5°; dihydrochloride, m.p. 103.5—107.0°; 125 (neutral) and 126 (acidic); triiodide, m.p. 126.5—128° (dihydrochloride); 126 derivative, m.p. 117—119°.

BIBLIOGRAPHIC LITERATURE CLASSIFICATION									
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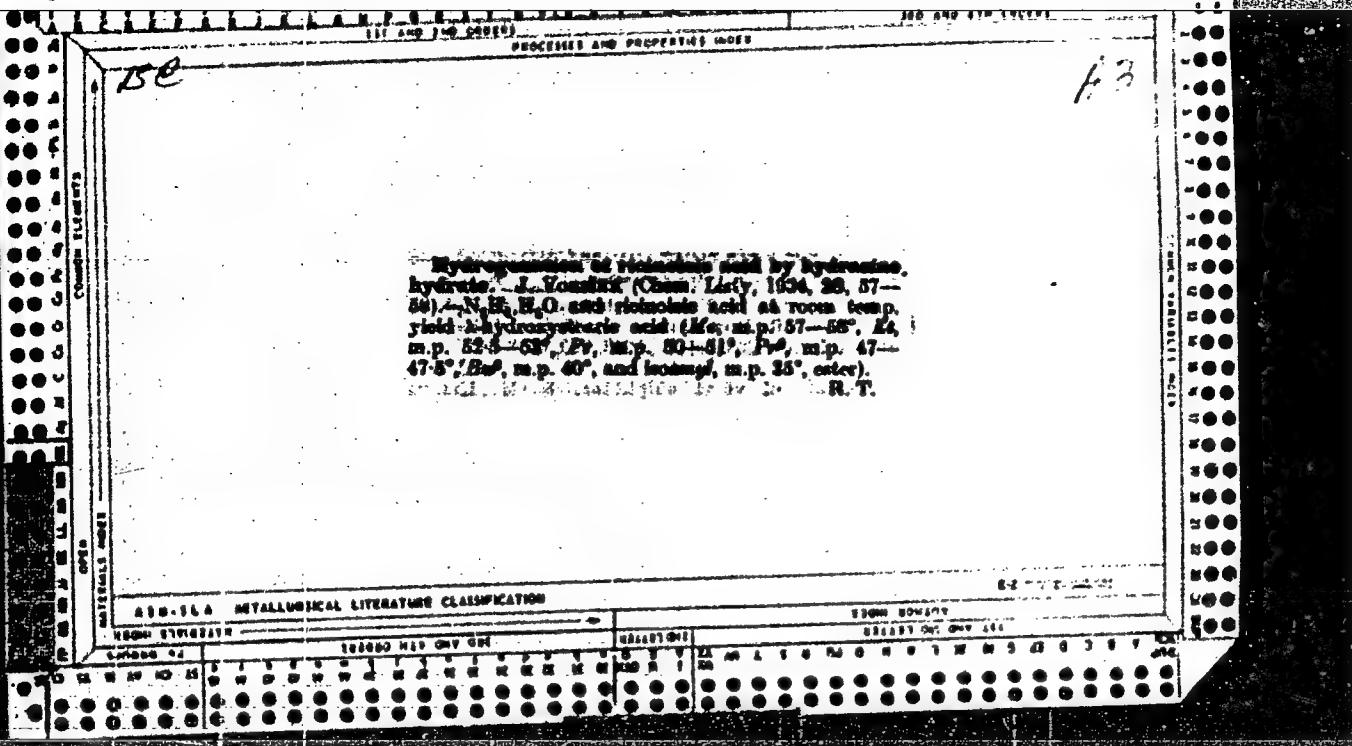
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by this part of the hydrochloric acid index. 1  
Voloshin (Chem. Listy, 1936, 22, 215-219) - 10 c.c. of  
conc. HCl are added to 100 c.c. of milk; 10 c.c. of the  
whey are evaporated to dryness; the residue is treated,  
the alk neutralized with conc. HCl, heated at 100° to  
eliminate NH<sub>3</sub>; HCl dissolved in HNO<sub>3</sub>, and Cl<sup>-</sup> deter-  
mined by titration with 0.1N-Hg(NO<sub>3</sub>)<sub>2</sub>. The no. of  
c.c. used is termed the HCl index (I<sub>H</sub>) and ranges from  
10 to 13.6 for normal milk. Values of > 16 occur  
immediately before or after parturition, and in disease,  
while values < 10 show the milk is diluted. When  
Na<sub>2</sub>CO<sub>3</sub> has been added, I<sub>H</sub> increased by 1.00 for every  
0.1% of Na<sub>2</sub>CO<sub>3</sub> present.

## ABSTRACT METALLURGICAL LITERATURE CLASSIFICATION

627-100-100-100

SEARCHED	INDEXED	SERIALIZED	FILED
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U.S.A. & M.	W.G.D. & C.R.	M.I.T. & C.R.	U.S.A. & M.



Separation of uranium from manganese, zinc, calcium, strontium, barium and magnesium by means of latex. I. estimate. V. Hovorka and J. Vojtěch. Collection Czechoslovak. Chem. Chemos., 11, 122-131 (1936). Latex of sodium formaldehyde sulphite with  $\text{UO}_4^{2-}$ ,  $\text{Ag}^+$ ,  $\text{Pb}^{2+}$ ,  $\text{Hg}^{2+}$ ,  $\text{Ca}^{2+}$ ,  $\text{Sr}^{2+}$ ,  $\text{NH}_4^+$  and  $\text{Cu}^{2+}$ . The reagent is 1% w/v in 50% alc. It is excellent for sepr.  $\text{UO}_4^{2-}$  from alkali and alk. earth ions. The soln. of nitrate, acetate or chloride counts 0-240 mg. of  $\text{UO}_4$  in 40-100 ml. is heated to boiling and then treated with 0-60 ml. of the reagent. Then, to buffer the soln., 8-16 ml. of a cold, 10% w/v, of NaOH is added and the soln. is allowed to stand for 1 hr. at room temp. It is then filtered, washed with hot water or, better, with a soln. contg. 25 ml. of the reagent soln. in 50 ml. of water, and boiled to 170°. The results of about 150 analyses showed that the method gives results which are fairly close to the truth.

W. T. H.

CH

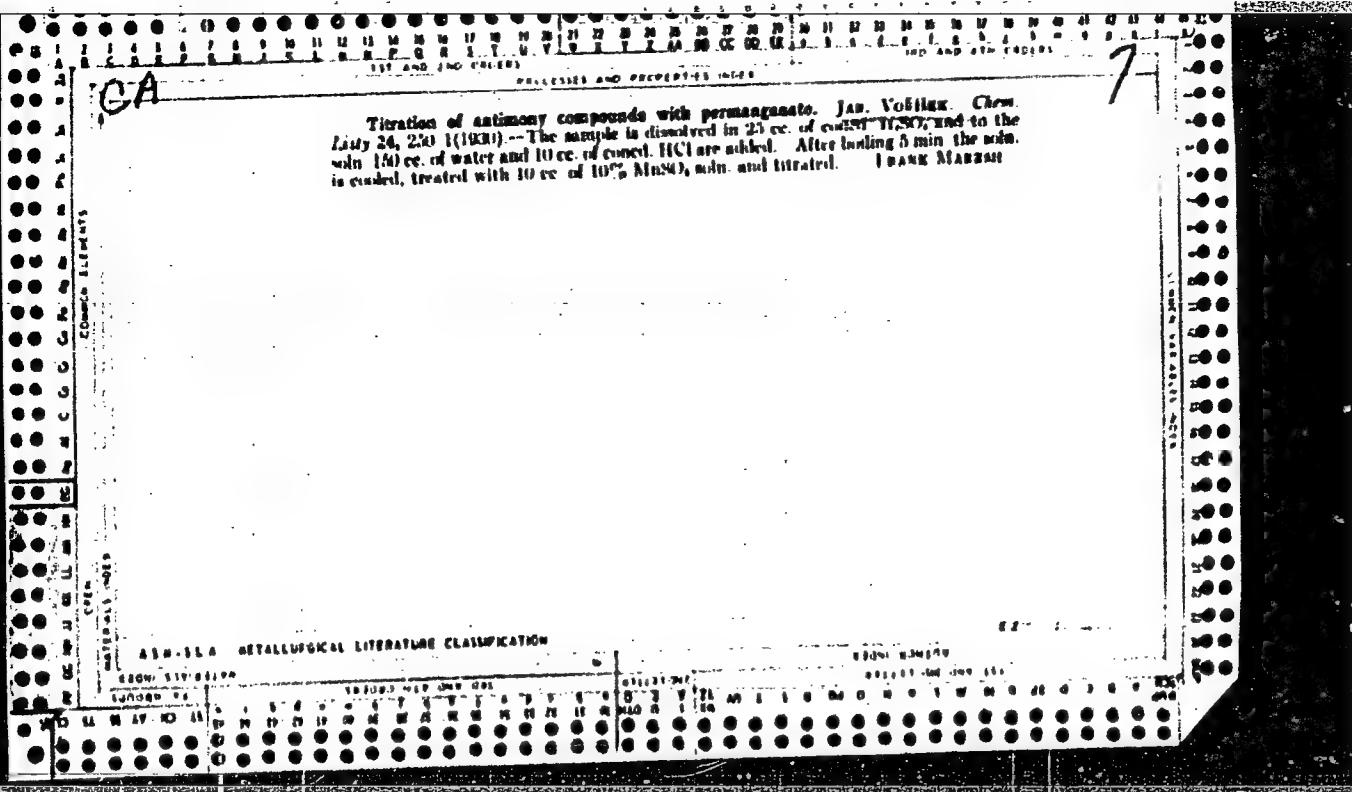
**Separation of Cu and Cd by Iodine-*J*-phenylpyrone.** V. Horváth and J. Vörösléb. *Chem. Listy* 37, 8-7 (1943); *Czech. Zeschr.* 1943, I, 1703-1; *C. A.* 37, 4211. — The method previously recommended for detg. Cu can be carried out in the presence of considerable Cu if sufficient NH<sub>4</sub> tartrate is present. Copper, Al, Fe, Pb, Ni, Co, etc., can be prevented likewise. To 200 slightly acidic soln. add about 8 ml. of 0.5 N H<sub>2</sub>SO<sub>4</sub>, 2.5-5.0 g. of NH<sub>4</sub> tartrate, DDI, to 100-150 ml., heat to 90° and add at once an excess of reagent (about 90 ml.) of a 1% soln. in water: MeOH = 1:2 for 0.1 g. Cu. After 3 hrs. filter, wash with 1-2% NH<sub>4</sub> tartrate soln., sprinkle with powdered oxalic acid and ignite to CuO. If the soln. is too acid at the start, neutralize with NH<sub>4</sub>OH. In the filtrate, Cd can be pptd. with (NH<sub>4</sub>)<sub>2</sub>S soln.

W. T. H.

7

CA

The hydrazide of *m*-nitrobenzoic acid as a new reagent for the determination of palladium. J. Vojtěk and Z. Vojtěk. *Chem. Listy* 37, 80-3, 65-70, 91-8(1943).—  
*m*-O<sub>2</sub>N-C<sub>6</sub>H<sub>4</sub>-CONHNH<sub>2</sub> (I) ppt. Pd quantitatively from acid soln. contg. HCl, H<sub>2</sub>SO<sub>4</sub>, and HNO<sub>3</sub>, forming (*m*-O<sub>2</sub>N-C<sub>6</sub>H<sub>4</sub>-CO-NHNH<sub>2</sub>)<sub>2</sub>PdCl<sub>4</sub> and the corresponding sulfate. After prolonged washing the chloride complex permits but the sulfate ions are substituted by hydroxyl. I ppt. Hg<sup>II</sup>, Cu, Fe, Ni, Au, Mo, Pd, Pt, and Os from neutral solns., but only Pd and Au from the acidic solns. The ppt. of Pd is yellowish, and is formed at the diln. 1:100,000 immediately; at the diln. 1:300,000 after 10 min. Add to the acid soln. 10-18 ml. of 1% Ba(OH)<sub>2</sub> soln. of I for 0.01 g. Pd, filter off the ppt. after gently heating, wash with 180-200 ml. of hot H<sub>2</sub>O, and ignite to Pd which is weighed. A procedure for sepp. Pd from other cations is given. Miks Hudlicky



The determination of sulfur dioxide as a preserving agent in dried fruit. Josef Hanau and Jar. Vodicka, Chem. Listy 31, 405-414 (1937).—The SO<sub>2</sub> content of apricots dried in ordinary ovens, and computed on a moist wt. remained const. (2341-2392 mg. SO<sub>2</sub> per kg.) for as long as 35 days; the SO<sub>2</sub> content began to decline only when the apricots ceased to lose any more water. This apparent constancy of the SO<sub>2</sub> content is due to the comparable loss of water, for when the SO<sub>2</sub> was lost, on the basis of dry matter, it dropped during the entire drying period (3382 to 2727 mg. of SO<sub>2</sub> in 30 days). When apricots which had been dried were remoistened and reduced, they continued to lose SO<sub>2</sub> during the 2nd drying process in amounts comparable to the loss during the 1st drying process. The culinary treatment of apricots (as a soaking in water) removed 26-33% of the original SO<sub>2</sub> and depended chiefly upon the vol. of water and less upon the time (1-4 hrs.). After such extd., apricots (containing 108 mg. of SO<sub>2</sub> per kg. of dry substance) were made into dumplings and were boiled; 2311 mg. of SO<sub>2</sub> remained in the apricots, 80 mg. of SO<sub>2</sub> diffused into the dough and 16 mg. of SO<sub>2</sub> volatilized into the air. Frank Maresch

## AMERICAN METALLURGICAL LITERATURE CLASSIFICATION

APPROVED FOR RELEASE: 03/14/2001

CIA-RDP86-00513R001860810014-2"

CA

5

The tetrathionate of hydrazine. Jan. Vohlan. Chem. Listy 26, 296-7 (1932). When hydrazine polyulfide (I) was mixed with oleic acid for 30 days in the presence of air, the oleic acid was hydrogenated to stearic acid and the S of I was oxidized to  $(\text{NH}_3)_2\text{SO}_4$  (II) forming colorless crystals, very sol. in  $\text{H}_2\text{O}$ , insol. in 40% EtOH or hot EtOH, stable in air, decomp. at 85-87°. It was also prep'd. by passing  $\text{SO}_2$  into I, heating until clear, filtering the poly-S and drying over anhyd.  $\text{CaCl}_2$ .

## AIA-SEA METALLURGICAL LITERATURE CLASSIFICATION

CA

5

The action of hydrazine polysulfide upon oleic acid. JAR. VOLLMER, Chem. Zeit. 24, 296-7 (1932).—Hydrazine polysulfide ( $(\text{NH}_2\text{NH}_2)_2\text{S}_x$ ) prep'd. by passing  $\text{H}_2\text{S}$  into hydrazine hydrate, was mixed with 10 g. oleic acid in the cold. The mist formed from the evolved  $\text{H}_2\text{S}$  and changes from a yellow to a colorless soln. in 20 days. After 30 days the solid mass was dissolved in warm 50% EtOH and treated with diluted HCl (1:1). After the ppt'd. fatty acid was washed with hot  $\text{H}_2\text{O}$ , dried and ext'd. with  $\text{CHCl}_3$ , the  $\text{CHCl}_3$ -sol. fraction was recrystl. 3 times from EtOH and  $\text{CHCl}_3$  and identified as stearic acid; yield 0.8 g., m. 100.8°. The  $\text{CHCl}_3$ -insol. fraction was recrystl. 3 times from EtOH and  $\text{CHCl}_3$  and identified as stearyl hydrazide; m. 114°, yield 2.8 g., N = 0.56%. A mist of oleic acid and hydrazine hydrate was refluxed over a free flame for 8 hrs while a const. stream of  $\text{H}_2\text{S}$  was passed through the mist. The product crystl. 3 times from EtOH was stearyl hydrazide, m. 113°; the filtrate contained oleic hydrazide. In a hot soln. oleic hydrazide forms and is hydrogenated in the presence of air to stearic hydrazide by a slight excess of  $(\text{NH}_2\text{NH}_2)_2\text{S}_x$ . FRANK MARSH

AIA-114 METALLURGICAL LITERATURE CLASSIFICATION

**Analysis of iron-nickel mixtures.** Josef Mareš and Jaroslav Vojtěk. *Chem. Listy* 29, 205-215 (1935).—For sptg. Fe from Ni the authors used and compared critically the acetate, succinate and hexamethylenetetramine methods. From a soln. having an Fe:Ni ratio larger than 40, the Fe was sptg. from Ni completely by a single pptn. in all 3 methods. When the Fe:Ni ratio exceeded 40, the Fe was sptg. from Ni by a double pptn. in the acetate and succinate method; in the hexamethylenetetramine method, the Fe pptd. still contained traces of Ni after 2 pptns. In order to prevent the adsorption of Ni upon the pptd. Fe during the sepn., the acidity of acetate soln. cannot exceed 3 cc. of *N* *AcH* per 500 cc. of soln. The adsorption of Ni upon the pptd. Fe is decreased by an addn. of *NH<sub>4</sub>Cl*. The ratio of Fe to Ni in the Fe ppt. is always const. for a given procedure. For the 2nd pptn., the Fe would be pptd. by the acetate or succinate method again and not by *NH<sub>4</sub>OH* which, when the Fe:Ni ratio exceeds 2, begins to give Fe pptn. contg. large quantities of Ni and demands a 3rd pptn. In slightly acid soln. contg. *NH<sub>4</sub>Cl* the acetate and succinate methods were equally accurate and useful. The results obtained by the hexamethylenetetramine method were inferior to those obtained with the acetate or succinate methods but remain better than those obtained by sptg. Fe from Ni by means of *NH<sub>4</sub>OH*.

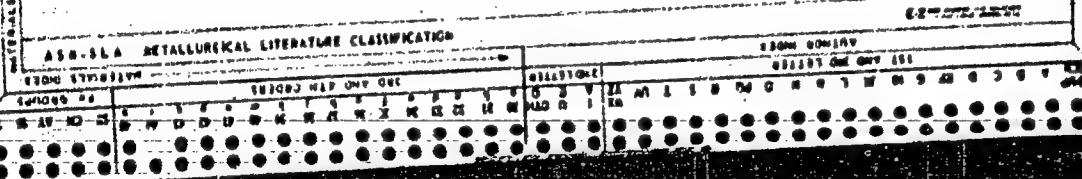
**APPROVED FOR RELEASE: 03/14/2001**

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ca

12

Determination of alkaline preservatives in milk by the hydrochloric acid number. Janssen-Veltink. Vestsil (Czechos. Akad. Zemedelského 10, 26-30/1a English 30) (1934).—The HCl no. is the no. of cc. of 0.1 N  $Hg(NO_3)_2$  necessary for titration of Cl bound by ash from 11 cc. of whey obtained from 100 cc. milk and pyrd. with 10 cc. concd. HCl. The HCl no. is theoretically increased 18.9 by the addn. of 1%  $Na_2CO_3$ . Add 10 cc. HCl (d. 1.19) to 100-cc. milk sample in a beaker, mix thoroughly, let stand 10 min., filter into a cylinder for sugar determination, return the first portion of whey over the filter. Pipet 10 cc. of whey into a 50-cc. porcelain dish, evap. to dryness on the water bath, dry at 130° and ash over a Bunsen burner, add concd. HCl, evap. on the water bath till all HCl disappears, dissolve in  $H_2O$ , add a few drops  $HNO_3$ , rinse into a beaker and titrate  $Cl^-$  with 0.1 N  $Hg(NO_3)_2$  with Na nitroprusside as indicator. The HCl no. in normal milk of one cow is 10.0-13.8. The amt. of  $Na_2CO_3$  added could be calcd. according to the formula: [(HCl no. - 11.0)/18.9] = 0.06%  $Na_2CO_3$ . If the HCl no. is less than 10, the milk has been watered. J. Kudera.



VORISEK, Jaroslav

Zero power heavy water reactor IR-0. Jaderna energie 9 no.8:  
264 Ag '63.

1. Ustav jaderneho vyzkumu, Ceskoslovenska akademie ved, Rez u  
Prahy.

(1)

CZECHOSLOVAKIA

LEJSEK, K., SEDLACEK, J., VORISEK, V. Chair of Chemistry and Pathological Physiology, Medical Faculty, Charles University, (Katedra Chemie a Patologické Fysiologie Lek. Fak. KU), Hradec Kralove.

"Oxygen Requirements of Lung and Liver Tissue After Diphosgene Poisoning."

Prague, Ceskoslovenska Fysiologie, Vol 15, No 2, Feb 66, p 77

Abstract: Experiments with sections of rabbit and frog organs were conducted. No difference of oxygen consumption due to poisoning was found; usage of glucose by the tissue did not change as a result of the poisoning. The lung parenchyma is heavily damaged by the poison. 1 Western, 1 East German, 1 Polish reference. Submitted at "16 Days of Physiology" at Kosice, 27 Sep 65.

1/1

VORISEK, M.

Vorisek, M. Star distribution caused by cosmic rays in nuclear emulsions. p.  
609. CESKOSLOVENSKY CASOPIS PRO FYZIKU. Praha. Vol. 4, no. 5, Oct. 1954.

SO: Monthly List of East European Accessions, (EEAL), LC, Vol. 4, No. 11,  
Nov. 1955, Uncl.

U. o RISEK, M.

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Reviewers: Bittner, Engle; Drury, Doctor; Smith; Tardieu; Simand;

**Engineer's Ed.** Stanislaw Voboff.  
**Engineer's Ed.** Stanislaw Voboff.  
**PURPOSE:** The book is intended for the general reader.  
**Coverage:** The book outlines the principles and operation of nuclear power plants and the use of radiosopes. The introductory chapters cover the fundamentals of nuclear physics and radioactivity. Several subsequent chapters deal with reactor physics, types of reactors, their engineering, control and

Instrumentation, operating and planned nuclear power instrumentation are described. A short chapter is devoted to the possibility of using nuclear power in transportation industry. The remaining chapters report on radiisotope gas industry, and on radiation hazards and safety measures, all and on radiology. There are 25 references. All personalities are mentioned.

CONTENTS

VORISEK, Miroslav

Determination of the moisture of materials by scattering neutrons on  
protons. Jaderna energie 3 no.9:258-271 S '57.

1. Ustav jaderne fysiky, Ceskoslovenska akademie ved, Praha.

~~SECRET~~, VCR 62-5, 101  
CZECHOSLOVAKIA/Nuclear Physics - Installations and Instruments.  
Methods of Measurement and Research

C-2

Abs Jour : Ref Zhur - Fizika, No 6, 1958, No 12460

Author : Vorisek Miroslav  
Inst : Institute of Nuclear Physics, Czechoslovak Academy of Sciences,  
Prague Czechoslovakia  
Title : Scintillation Detector for Slow Neutrons

Orig Pub : Ceskosl. casop. fys. 1957, 7, No 4, 396-407

Abstract : A mixture of ZnS (Ag) and  $B_2O_3$  is investigated with an aim toward using it for detection of thermal and resonant neutrons. The ratio of the ZnS (Ag): B ranging from 8:1 to 12:1 is optimum from the point of view both of the efficiency with respect to neutrons, and of the form of the integral spectrum. The best thickness of the layer of the mixture is 0.75 to 1.0 mm. For these optimum values, the efficiency of the mixture with respect to neutrons is 25% at a low background of gamma rays. At a strong background of gamma rays, it is possible to reduce the background to  $10^{-7}$ % with the aid of a discriminator, and the efficiency for neutrons remains not less than 5%.

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CZECHOSLOVAKIA/Nuclear Physics - Installations and Instruments.  
Methods of Measurement and Research

C-2

Abs Jour : Ref Zhur - Fizika, No 11, 1958, No 24543

Author : Vorisek Miroslav  
Inst : Institute of Nuclear Physics, Prague, Czechoslovakia  
Title : Detector for Slow Neutrons.

Orig Pub : Czechosl. fiz. zh., 1957, 7, No 6, 757-766

Abstract : A scintillation counter is proposed for slow neutrons. The counter is obtained by sintering ZnS (Ag) and  $B_2O_3$ . The experimentally-obtained optimum value of the ratio  $ZnS (Ag)/B$ , which is approximately equal to 9:1, makes it possible to obtain a scintillator of large dimensions with a thickness from 0.75 to 1 mm with sufficiently high efficiency with respect to registration of thermal neutrons (25%). The detector can operate with a strong background of gamma rays because of its different sensitivity to gamma rays (from  $10^{-4}$  to  $10^{-8}$ %) and thermal neutrons (from 10 to 5% respectively). The counter has a resolution on the order of  $10^{-7}$  seconds and a sufficiently good stability.

Card : 1/1

VORISEK, M.

AUTHOR: Miroslav Vorisek

CZECH/37-59-2-6/20

TITLE: The Absorption of a Beam of Neutrons in Absorbers of Different Shapes

PERIODICAL: Československý Časopis Pro Fysiku, 1959, Nr 2,  
pp 157-166

ABSTRACT: The present paper limits itself to the calculation of absorption of a beam of neutrons in those cases when the effective cross-section for absorption varies as the reciprocal of the velocity. The calculation leads, in most cases, to integrals which cannot be expressed by elementary functions. They can, however, be evaluated by special functions or rapidly converging series. This is often quicker than numerical integration. The present work sets out to supplement the known results for the absorption of a beam of neutrons in a plate (Ref 1) by including the absorption of mono-energetic neutrons (first part), thermal neutrons (second part) and resonance neutrons (third part), in absorbers of spherical shape (either full or hollow) and of cylindrical shape - again full or hollow. The exact solutions, as well as approximate solutions, are discussed. The following

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CZECH/37-59-2-6/20

The Absorption of a Beam of Neutrons in Absorbers of Different Shapes

assumptions are made: a) the absorption of mono-energetic neutrons is governed by an exponential law (Ref 2); b) the scattering is small compared with the absorption; c) the size of the absorber is assumed small in the direction of the beam, compared with the mean free path of scattering. Under these assumptions, the number of neutrons passed through the absorber in unit-time is given by Eq (1). The absorption is given by the difference between the number of incident neutrons and equation (1), i.e. Eq (3). From Eqs (1) and (3) the probability of absorption and transmission is calculated (Eq (4)). Eqs (5) - (7) indicate means of solving Eq (4), while Eqs (8), (9) and (10) show approximate solutions.

Part I. (Parts 2 and 3 are being prepared).

The absorption of mono-energetic neutrons in a plate is given by the well-known exponential law (Eq (12)).

Eq (15) gives the numbers of neutrons transmitted through a sphere per unit-time. Eq (15) can be transformed to (16) and an approximate solution found by Eq (17). ✓

Card 2/3

CZECH/37-59-2-6/20

The Absorption of a Beam of Neutrons in Absorbers of Different Shapes

Eqs (18)-(24) deal with absorption in a hollow sphere.

Eqs (25)-(30) deal with absorption in a cylinder.

Eqs (31)-(47) deal with absorption in a hollow cylinder.

There are 4 tables and 7 references, of which 6 are English and 1 is Soviet.

ASSOCIATION: Ustav jaderneho vyzkumu CSAV, Praha  
(Institute of Nuclear Physics, Ac. Sc., Prague)

SUBMITTED: September 16, 1958

Card 3/3

✓

VORISEK, Miroslav

Distribution of thermal neutron absorption density in  
fuel cells from natural uranium. Jaderna energie 10 no.11:  
407 N '64.

1. Institute of Nuclear Research of the Czechoslovak Academy  
of Sciences, near Prague.

VORISEK, Miroslav

Distribution of the density of thermal neutron absorption in  
fuel elements with internal structure. Jaderna energie 9 no.8:  
264 Ag '63.

1. Ustav jaderneho vyzkumu, Ceskoslovenska akademie ved, Rez  
u Prahy.

VORISEK, M.

International symposium of the International Atomic Energy Agency  
on exponential and critical series. Jaderna energie 10 no. 3:  
106-107 Mr '64.

ACC NR: AP7002326

SOURCE CODE: C2/0038/66/000/005/0101/0107

AUTHOR: Chochlovsky, Igor--Khodlowski, I.; Riha, Karol--Rzhiga, K.; Panyr, Milos; Vorisek, Miroslav--Vorzhishok, M.; Charrad, Brotislav--Kharad, B.

ORG: Chochlovsky; Riha; Panyr /Chonoprojekt, Prague; Vorisek; Charrad/ Instituto  
of Nuclear Research, CSN, Roz (Ustav jadernho vyzkumu CSAV)TITLE: TR-0 heavy water zero-power reactor of Nuclear Research Institute of  
Czechoslovakian Academy of Sciences

SOURCE: Jaderna energie, no. 5, 1966, 161-165

TOPIC TAGS: research reactor, heavy water

ABSTRACT: The zero-power heavy water reactor TR-0, a pulsed neutron source and an exponential heavy water system, is described. This reactor has rod-shaped fuel elements of natural uranium. The active zone has a diameter of 3500 mm and a height up to 4000 mm. Its auxiliary layout was selected so that long-term studies on heavy water reactor lattices could be carried out. The principles of the long-term experimental program are outlined. The engineering solutions with respect to the reactor vessel and its system for the automatic adjustment of the lattice support and to the reactor circuits are described. The principal circuits considered are the heavy water circuit and the inert gas circuit in which dry air is used. A brief description is given of the construction work. This article was presented by F. Klik. Orig. art. has: 2 figures and 6 tables. [NA]

SUB CODE: 18 / SUBM DATE: 14Oct65

UDC: 621.039.5TR-0 621.039.524.46 621.039.5(437)

Card 1/1

VORISKOVA, M.; Technicka spoluprace: OBSILOVA, F.

Diagnostic value of the amyl nitrite test. Cesk. pediat. 20  
no.8:693-698 Ag '65.

1. II. detska klinika fakulty detskeho lekarstvi Karlovy  
University v Praze (prednosta prof. dr. J. Houstek, DrSc.).

VORISEK, P.

Experimental and clinical problems on the influence of ionizing  
radiations on the development of the fetus. Cas. lek. cesk. 101  
no. 50: 241-246 14 D '62.

1. Ustav pro paci o matku a dite v Praze-Podoli, reditel doc.

dr. M. Vojta.

(FETUS) (EMBRYO) (RADIATION INJURY)  
(ABNORMALITIES)

VORISEK, P.

Effect of small doses of ionizing radiations on the ovary and on its function. Cas.lek.cesk 100 no.42 Lek veda zahr:217-224 20 0 '61.

1. Ustav pro peci o matku a dite, Praha-Podoli, reditel doc. MUDr. M. Vojta, zaslouzilý lekar CSSR.

(OVARIES radiation eff)

VORISEK, Premysl (Czechoslovakia)

Cable spinning of cotton industry synthetic fibers. Magy  
textil 1' no.1:25-28 Ja '65.

EXCERPTA MEDICA Sec 8 Vol 12/12 Neurology Dec 59

6279. RESERPINE TREATMENT OF HUNTINGTON'S CHOREA AND OTHER EXTRAPYRAMIDAL SYNDROMES - Reserpine v léčbě Huntingtonovy chorey a některých jiných extrapyramido-vých syndromů - Vorisek V. *Neurol. Odd. Nemocnice na Bulovce, Praha - ČSL.* NEUROL. 1958, 21/2 (99-105)

An account of 2 cases of Huntington's chorea treated with reserpine. One of the cases of average severity has been under observation for over one year, the second in an advanced condition for almost half a year. In both cases favourable results were achieved affecting both the choreatic movements and the mental disturbances. In 2 additional cases of Huntington's chorea, which were under observation for a shorter period, good results have also been achieved. These results were obtained by using relatively small doses regularly. No side effects were noticed. In 2 cases of athetosis treated with reserpine the results were much less favourable; the quantity of reserpine used was also larger. In one case of organic tic there was no improvement after reserpine. (VIII, 14\*)

VORISEK, V.

Information on static resistance of welded joints in carbon-poor steel.  
p. 345. ZVARANIE. (Ministerstvo hutneho prumyslu a rudnych bini a  
Ministerstvo strojarstva) Bratislava. Vol. 3, no. 11, Nov. 1954.

SOURCE: East European Acquisitions List, Vol. 5, no. 9, September 1956

VACHA, Karel; VORISEK, Vladimir; CHROBAK, Ladislav

Significance of detecting nucleated erythrocytes in the peripheral blood. Sborn. ved. prac. lek. fak. Karlov. Univ. (Hrad. Kral.) 6 no.4:435-442 '63.

1. I. interni klinika; prednosta: prof. MUDr. F.Cernik.

\*

VORISEK, Vladimir, inz. CSc.

Real effect of prestressed anchored poles for very high voltage  
lines. Inz stavby 12 no. 71308-315 J1'64

1. Slovak Higher School of Technology, Chair of Metal and Wood  
Constructions.

CZECHOSLOVAKIA / Pharmacology, Toxicology. Tranquillizers.

Abs Jour: Ref Zhur-Biol., No 18, 1958, 85088.

Author : Vorisek, Vlastimil,

Inst : Not given.

Title : The Treatment with Reserpine of Huntington's Chorea  
and of Other Syndromes of Extrapyramidal Dysfunction.

Orig Pub: Ceskosl. neurol., 1958, Vol 21, No 2, 99-105.

Abstract: Description is given of good results obtained in  
the treatment, with comparatively small doses of  
reserpine (R), of four patients with Huntington's  
chorea. R influenced both the hyperkinesia and  
the psychic disorders. No side effects were noted.  
In two patients with athetosis, treated with larger  
doses, R was less effective. In organic tic, no  
improvement was seen. -- Yu. R.

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111 AND 2000 COPIES  
PROCESSED AND PROPERTIES INDEX

4

Adsorption effect on the polarographic curves of pyrocyanine. M. Vuríčková (Balovka Hloub., Prague). Collection Czechoslov. Chem. Commun. 12, 107-10 (1947) (in English); cf. Müller, C.A. 37, KNP 39, 30349. -- The polarographic reduction of pyrocyanine ( $\alpha$ -hydroxy-methylphenazine) (I) was investigated over the pH range 11-12. At concns. of I up to  $10^{-4}$  M in alk. solns. reduction proceeds in a single wave which shows a more pos. potential than the corresponding potentiometric value. At higher concns. 2 waves of unequal height appear, the first being always higher by a value that is independent of the concn. In alk. solns. up to a pH of 10 an extra wave appears at more pos. potentials the height of which is likewise independent of the concn. of I. These anomalies are due to the adsorption of undissolved mols. of dihydropyrocyanine according to Brdička's theory (cf. C.A. 37, 6601\*). G. Reed

C 1)

Polarographic studies with dropping-mercury electrode.  
LXXXIV. Separation and reduction potentials of metallic  
ions in ammoniacal solutions. M. Volkova. Collection  
Chem. Chem. 11, No. 6(1959). The  
reduction potentials of a series of ions were studied in 10  
molar, 1 N with NH<sub>3</sub> and 1 A with NH<sub>4</sub>Cl. The studies  
were made at 18° with a dropping-mercury electrode. The  
half-wave potentials obtained were Cu<sup>+</sup> -0.273, Cu<sup>0</sup> -0.359,  
Tl<sup>+</sup> -0.519, Cu<sup>2+</sup> -0.538, Cu<sup>3+</sup> -0.862, Ni<sup>2+</sup>  
-1.135, VO<sub>2</sub> -1.231, Cu<sup>4+</sup> -1.320, Zn<sup>2+</sup> -1.381, Cr<sup>6+</sup>  
-1.46, Fe<sup>2+</sup> -1.618, Mn<sup>2+</sup> -1.688, Cr<sup>3+</sup> -1.74 v. The  
reduction process for vanadate is not reversible. Addition  
of small amounts of gelatin shifts it to a neg. value.  
Clarice E. Hickey

VORISOV Yu. P.

AFANASIEVA, A.Y., BAISHEV, B.T., VORISOV, YU.P., VASILYeva, V.N.,  
VOLOMOV, V.V., ZINOVIEVA, L.A., KAMENETSKIY, S.D., MAKISOV, M.I.,  
MAKISOV, M.H., MATDEBOR, V.N., NOVIKOV, I.P., SOKOLOVSKII, N.V.,  
SUSHILIN, V.A., YAKOVLEV, V.P.

Problem of developing oil in the USSR

Report to be submitted for the Sixth World Petroleum Congress  
Frankfurt, 16-26 June 63

VORISOV, Yu.Ya., and MASHKOVA, T.I.

"Experimental work on the acceleration of drying in an acoustic field."

Report presented at the All-Union Scientific-Engineering Conference on  
the Application of Ultrasonics in Industry, Moscow, 22-26 November 1960.

VORK, Hnas, prof.; POBUL, G., kand. tekhn. nauk, retsenzent; ABO, L.,  
red.; TIMMER, K., tekhn. red.

[Steel overhead lines] Ohuliinid terasjuhtmeist. Teine, ümber-  
tootatud trukk. Tallinn, Eesti riiklik kirjastus, 1961. 78 p.  
(MIRA 15:5)

(Electric lines—Overhead)

DUMAYEVSKIY, M.M.; IL'INSKIY, B.D.; SINEBRYUKHOV, N.V.; VORKEL', M.M.;  
ZORIN, S.V., red.; DOBUZHINSKAYA, L.V., tekhn.red.

[Safety regulations in rolling-mill practice] Pravila bez-  
opasnosti v prokatnom proizvodstve. Moskva, Gos.nauchno-tekhnik.  
izd-vo lit-ry po chernoi i tsvetnoi metallurgii, 1960. 112 p.  
(MIRA 13:7)

1. Soyuz rabochikh metallurgicheskoy promyshlennosti. Tsentral'-nyy komitet. 2. Vsesoyuznyy nauchno-issledovatel'skiy institut organizatsii proizvodstva i truda chernoy metallurgii (for Dumayevskiy, Il'inskiy, Sinebryukhov, Vorkel').  
(Rolling mills--Safety measures)

FOJTIK, Frantisek; TOUPALOVA, Hana; VORISEK, Vlastimil

Artificial hibernation in severe cranial & brain injuries. Cas.  
lek. cesk. 97 no.30:927-932 18 July 58.

1. Chirurgicka klinicka zakladna UDL, prednosta prof. MUDr. Jan  
Knobloch, neurologicka oddeleni, prednosta prof. MUDr. Otakar Janota,  
v Praze 8-na Bulovce. F. F. Praha 8, Nad Rokoskou 21.

(BRAIN, wds. & inj.

ther., artif. hibernation (Cz))

(HIBERNATION, ARTIFICIAL, in var dis.  
craniocerebral inj. (Cz))

VORISEK, Vlastimil

Reserpine treatment of Huntington's chorea & other extrapyramidal syndromes. Cesk. neur. 21 no. 2:99-105 Mar 58.

1. Neurologické oddelení nemocnice na Bulovce v Praze 8, prednosta prof. Dr Otakar Janota.

(HUNTINGTON'S CHOREA, ther.

reserpine (Cz))

(EXTRAPYRAMIDAL TRACT, dis.

ther., reserpine (Cz))

(RESERPINE, ther. use

Huntington's chorea & other extrapyramidal disord. (Cz))

VORISEK, V.

"Experimental contribution to the problem of the function of microelements in plant nutrition. IV. Experiments with the potato (*Solanum lycopersicum*). II." Chemike Zvesti, Bratislava, Vol 6, No 3/4, Mar./Apr. 1952, p. 299

SO: Eastern European Accessions List, Vol 3, No 10, Oct 1954, Lib. of Congress

"APPROVED FOR RELEASE: 03/14/2001

CIA-RDP86-00513R001860810014-2

VORISERK MIRROSI HV

APPROVED FOR RELEASE: 03/14/2001

CIA-RDP86-00513R001860810014-2"

VORISOV, E.

Title: Elimination of the radio-receiving interferences produced by the  
ST-35 apparatus

Author: E. Vorisov

Publication: Red Army Communications

No. 2-3, p. 36-39 Date: 1944

From List ATIC 20361-1

JANES, Hans; KAASIK, Paul; PUUSEPP, Eugen; VOLDEK, Aleksander; VORK, H.,  
prof., retsenzent; OORN, F., inzh., retsenzent; ABO, L., red.;  
VAHTRE, I., tekhn. red.

[Electric machinery] Elektrimasinad [By] H.Janes ja teised.  
Tallinn, Eesti riiklik kirjastus, 1961. 647 p. (MIRA 15:5)  
(Electric generators) (Electric transformers)

USSR/General Problems.

A-

Abs Jour : Ref Zhur - Khimiya, No 10, 1957, 33422

Author : Vork, Z.K., Ivanchenko, A.S.  
Inst :

Title : Electrolyzer with a Coal Screen.

Orig Pub : Khimiya v shkole, 1957, No 1, 63-64.

Abstract : A scheme and the description of the apparatus is given.  
Instructions for carrying out the experiments are also included.

Card 1/1

VORKACHEV, G. G. Cand. Agric. Sci.

"Some Results of the First Year of Reclamation of Virgin and Fallow Lands  
in the Altay," Agrobiol., No.3, pp. 106-110, 1955

Altay Zonal Scientific Research Inst. of "griculture and Animal Husbandry, Barnaul  
Translation 2030158

USSR/Cultivated Plants - Technical, Oleaginous, Sacchariferous.

II-7

Abs Jour : Ref Zhur - Biol., No 3, 1950, 39421

Author : Vorkachev, G.G.

Inst : All-Union Scientific Research Institute of East Crops.

Title : Efficiency of Fertilizers Used While Planting Southern  
Kemp Under Conditions Prevailing in Northern Caucasus.

Orig Pub : Tr. Vses. N.-i. in-t lub. kul'tur, 1957, vyp. 22, 85-86.

Abstract : No abstract.

Card 1/1

DMITRIYEVSKIY, K.I., master-vzryvnik; BYCHKOV, F.; NIKITIN, L., inzh.;  
VORKHLIK, M., inzh.; TYUTRIN, V., inzh.; YUDINA, N.F., inzh.;  
ZANEGIN, U., inzh.

Editor's mail. Bezop. truda v prom. 5 no.8:34 Ag '61.  
(MIRA 14:8)

1. Shakta No.32, Stalinskaya oblast' (for Dmitriyevskiy).
2. Sherlovozorskij gornoobogatitel'nyy kombinat, Chitinskaya oblast'  
(for Nikitin, Vorkhlik, Tyutrin). 3. Otdel tekhniki bezopasnosti  
Nizhne-Tagil'skogo metallurgicheskogo kombinata imeni V.I. Lenina  
(for Yudina). 4. Tekhnicheskiy otdel tresta Dorogobuzhshakhtostroy  
(for Zanegin).

(Mining engineering--Safety measures)

VORKHOBOV, L.A.

Treatment of abscesses and fistulae in children after the injection  
of medical substances. Sov.med. 25 no.8:94-97 Ag '60.

(MIRA 13:9)

1. Iz detskoy klinicheskoy bol'nitsy im. N.F.Filatova (glavnnyy vrach  
M.N.Kalugina) i kliniki detskoy khirurgii (zav. kafedroy - prof.  
S.D. Ternovskiy) II Moskovskogo meditsinskogo instituta imeni  
N.I. Pirogova.

(ABSCESS)

(FISTULA)

(INJECTIONS)

VOJKAPIC, M.; RADENOVIC, M.

Supplying ammunition to advanced detachments in defense, p. 25

VOJNI GLASNIK (Jugoslavenska narodna armija) Beograd, Yugoslavia.  
Vol. 13, no. 1, Jan 1959

Monthly List of East European Accessions EEAI LC, Vol. 8, no. 6, June 1959  
Uncla.

VORKEL', M.M.

Industrial traumatism in rolling mills. Metallurg 7 no.3:35-36  
Mr '61.  
(Rolling mills--Safety measures)

VORKOV, Sergey Stepanovich, kontradmiral; POLIKARPOV, V.D., red.  
BUKOVSKAYA, N.A., tekhn. red.

[Flag on the gaff] Flag na gafele. Moskva, Voenizdat, 1962.  
(MIRA 15:7)  
127 p.  
(Black Sea--World War, 1939-1945—Personal narratives)

VORKOV, Yu., gvardii podpolkovnik, voyennyy letchik pervogo klassa;  
NEDEL'KIN, V., kapitan

Radar determination of wind. Av.i kosm. 46 no.9:46-47 S '63.  
(MIRA 16:10)

VORKOVASTOV, K.S., gornyy inzhener-marksheyder

Profiling vertical mine workings. Gor. zhur. no.3:50-52 Mr. '63.  
(MIRÁ 16:4)

1. Magadanskiy okrug Gosudarstvennogo komiteta pri Sovete Ministrov  
RSFSR po nadzoru za bezopasnym vedeniem rabot v promyshlennosti i  
gornomu nadzoru.

RODIONOV, L. Ye., kand. tekhn. nauk; VORKOVASTOV, K. S., gornyy inzh.

Accuracy of a mine survey in working placer deposits by the  
open-pit method. Gor. zhur. no.11:64-67 N '62.  
(MIRA 15:10)

1. Vsesoyuznyy zaochnyy politekhnicheskiy institut, Moskva  
(for Rodionov). 2. Magadanskiy sovet narodnogo khozyaystva  
(for Vorkovastov).

(Mine surveying)

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CIA-RDP86-00513R001860810014-2

VORKUL', M.L., inzh.; ISKENDEROV, I.M., inzh.

Machinery for working rock. Stroi. i dor. mash. 9 no. 7:12-14 51 '64.  
(MIRA 18:3)

APPROVED FOR RELEASE: 03/14/2001

CIA-RDP86-00513R001860810014-2"

BUKRINSKAYA, A.O.; GITEL'MAN, A.K.; VORKUNOVA, O.K.

Early proteins of myxoviruses. Vop. virus. 9 no. 5: 569-575  
(MIRA 18:6)  
S-O '64.

1. Institut virusologii imeni Ivanovskogo AMN SSSR, Moskva.

BUKRINSKAYA, A.G.; VORKUNOVA, G.K.

Reproduction of ribonucleic acid of the influenza virus in  
the presence of low concentrations of actinomycin D. Vop.  
virus. 9 no.6:657-661 N-D '64.

(MIRA 18:11)

1. Institut virusologii imeni D.I. Ivanovskogo AMN SSSR, Moskva.

"APPROVED FOR RELEASE: 03/14/2001

CIA-RDP86-00513R001860810014-2

BUKRINSKAYA, A. G.; AZADOVA, N. B.; GIVEL'MAN, A. K.; VORKUNOVA, G. K.

"Nekotorye zakonomernosti reproduktsii rnk-miksovirusov."

report presented at Symp on Virus Diseases, Moscow, 6-9 Oct 64.

Institut virusologii im D. I. Ivanovskogo AMN SSSR, Moskva.

APPROVED FOR RELEASE: 03/14/2001

CIA-RDP86-00513R001860810014-2"

VORKUT, A., inzh.

Using a device in scheduling daily shift assignments. Avt. transp.  
42 no. 9:19-21 S '64. (MIRA 17:II)

1. Kiyavskiy avtomobil'no-dorozhnyy institut.

VORLICEK, I.

"Dynamics of the impulse-type controller with variable time rate and frequency of impulses."

Automatisace. Praha, Czechoslovakia. Vol. 2, no. 3, Mar. 1959.

Monthly list of East European Accessions (EEAI), IC, Vol. 8, No. 6, Jun 59, Unclassified

"APPROVED FOR RELEASE: 03/14/2001

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APPROVED FOR RELEASE: 03/14/2001

CIA-RDP86-00513R001860810014-2"

S/271/63/000/001/020/047  
D413/D308

AUTHOR: Vorlichek, Ivo

TITLE: An extremal regulator

PERIODICAL: Referativnyy zhurnal, Avtomatika, telemekhanika i vychislitel'naya tekhnika, no. 1, 1963, 49, abstract 1A270 (Czech pat., cl. 21 c, 46/51, no. 99848, June 15, 1961)

TEXT: The patent covers a regulator of extremal type, which serves for an automatic adjustment of a controlled quantity to its optimal value (maximum or minimum). The device consists of a two-position impulse regulator activated by the difference between the signals from the controlled quantity and from an element of the regulator which is adjusted manually. The regulator is connected in a circuit with two stable states that controls a two-position switch or commutator and also an element acting on the controlled quantity (a servomotor). To set up for the minimum or maximum, the manual regulator signal is adjusted to a value slightly higher than the

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An extremal regulator

S/271/63/000/001/020/047  
D413/D308

maximum or lower than the minimum value of the controlled quantity permitted by production conditions.

[Abstracter's note: Complete translation]

Card 2/2

CZECHOSLOVAKIA

VORLICEK, J., VYBRA, P.

Research Institute ZIM, Mnisek pod Brdy, and J.Heyrovsky Institute of  
Polarography, Czechoslovak Academy of Sciences, - Prague - (for both).

Prague, Collection of Czechoslovak Chemical Communications, No 12,  
December 1965, pp 4272-4279

"Amperometry with two polarisable electrodes. Part I: Chelometric  
determination of iron (3) with Pt-Pt electrode system indication."  
(To the 75th birthday of Academician J.Heyrovsky).

CZECHOSLOVAKIA

VYTRA, F; VORLICK, J.

1. J.Heyrovsky Institute of Polarography, Czechoslovak Academy  
of Sciences, Prague - (for ?); 2. Research Institute of Iron  
Ore Mines, Mnisek pod Brdy - (for ?)

Prague, Collection of Czechoslovak Chemical Communications,  
No 1, January 1966, pp 51-57

"Amperometry with two polarizable electrodes. Part 4: Direct  
chelometric determination of thorium."

VORLICEK, Jan, RNDr.; DOSTAL, Jan

Determining carbon in graphite raw materials and concentrates.  
Rudy 12 no.6:181-182 Je '64.

1. Research Institute of the Zelozorudne doly a hrudkovny,  
Mnisek pod Brdy.

VORLICEK, J., RNDr.; DOLEZAL, J., doc., dr.

Fast titration determination of antimony in ores and concentrates.  
Hut listy 18 no.1:55-56 Ja '63.

1. Vyzkumný ustav, Zálesné doly a hrudkovny, Mnisek pod Brdy  
(for Vorlicek). 2. Katedra analytické chemie, Karlova  
univerzita, Praha (for Dolezal).

VORLICEK, Jan, RNDr.; HAVLICEK, Vaclav

Titration determination of carbon dioxide in ores.

Rudy 11 no.3:87-88 Mr '63.

1. Vyzkumny ustav zolezorudnych dolu a hrudkoven, Mnisek pod  
Brdy.

VORLICEK, J.

VORLICEK, J.; SEFERKA, I.

"Study on Corrosion. I. Contribution to the Polarographic Study on the  
Corrosion of Metals", p. 920, (CHMICKE LISTY, Vol. 48, No. 6, June 1954,  
Praha, Czech.)

SO: Monthly List of East European Accessions (EEAL), LC, Vol. 4, No. 3,  
March 1955, Uncl.

VORLICEK, Jan, RNDr.; VYDRA, Frantisek, inz., CSc.

Direct complexometric determination of  $\text{Fe}^{3+}$  in ores. Hut listy  
18 no.10:733-734 O '63.

1. Vyzkumny ustav zelozorudnych dolu a hrudkoven, Mnisek pod Brdy  
(for Vorlicek). 2. Polarograficky ustav, Ceskoslovenska akademie  
ved, Praha (for Vydra).

Distr: ME2c

V. Corrosion studies. XVIII. Processes governing the kinetics of dissolving of metal. Ivan Šekera, Karel Smrk, Jan Vodrážek, and Eduard Beránek (Výzkumný ústav ochrany materiálů G. V. Akinzová, Prague). Chem. Listy 52, 1206-17 (1958); cf. C.A. 52, 19811e. — The dissolving of metals in acids or bases may be controlled by 2 steps according to the concn.: up to the concn. 0.1*N* the rate is controlled by the diffusion of H<sup>+</sup> ions to the metal surface; at concns. greater than 0.5*N* the rate controlling step is the discharging of H<sup>+</sup> ions, and in the range from 0.1 up to 0.5*N* the dissolving action is controlled by both steps. Activation energies for some metals and media were detd. in all 3 ranges mentioned. XIX. Kinetics of dissolving of metal. Karel Šarček, Ivan Šekera, Jaroslav Prášek, Eduard Beránek, and Jan Vodrážek. Ibid., 1312-17. — The time dependence and temp. dependence of the dissolving rate of metals in aq. solns. at const. concn. of the aggressive component was detd. in cases where no insol. reaction products are formed on the metal surface. The kinetic equation is of the zeroth order. The results are expressed by an empirical equation in the form:  $\log K = a_1 \exp(a_2 c) - a_3 T^{-1} \exp(a_4 c) + \log t - \gamma$ , where *K* is the amt. of the metal dissolved in the time *t*, at abs. temp. *T*, and *c* is the concn. of the soln. The applicability range of this equation is discussed. XX. Effect of light on the kinetics of corrosion processes. Ibid., 1218-21. — Light accelerates the corrosion process in which no layers of corrosion products are formed on the metal surface. Light energy increases the rates of the process (both cathodic and anodic) but does not change its mechanism. G. Erdélyi

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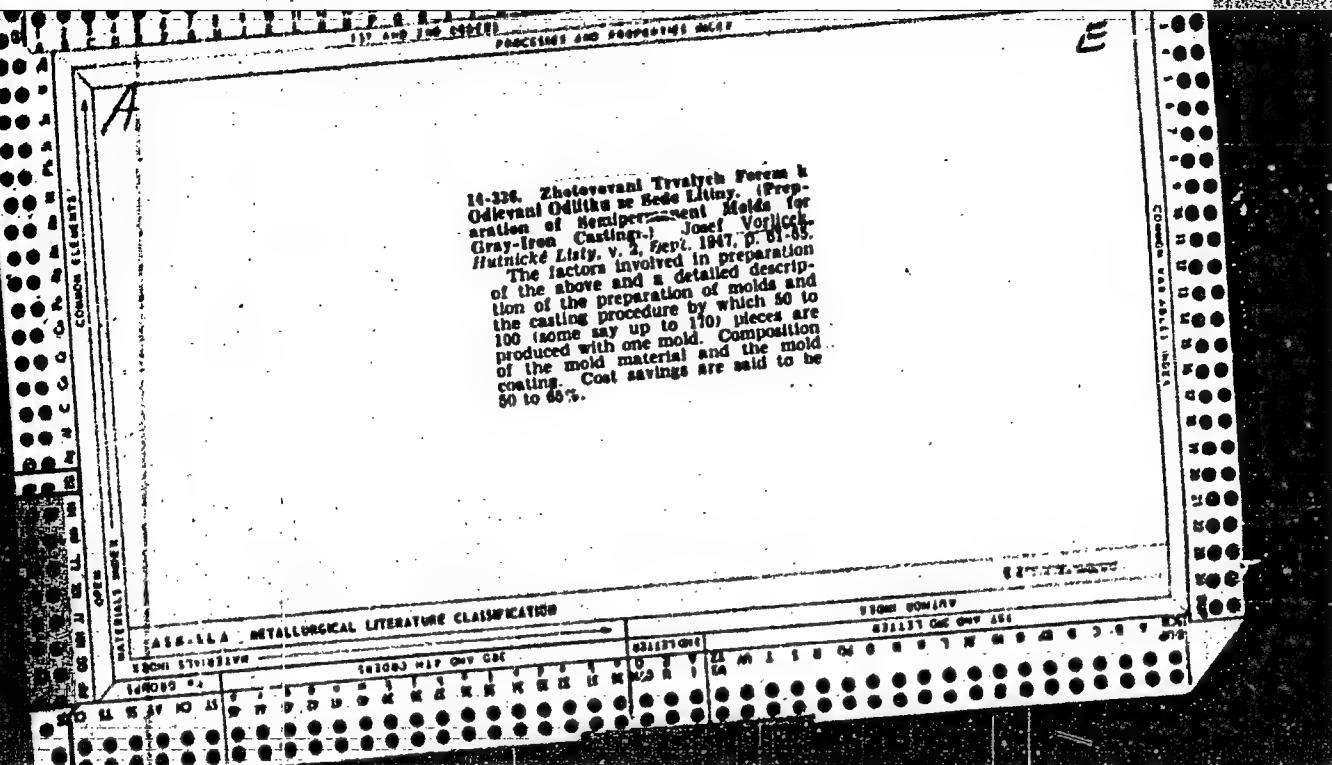
APPROVED FOR RELEASE: 03/14/2001

CIA-RDP86-00513R001860810014-2"

VORLICEK, J., SEKERKA, I.

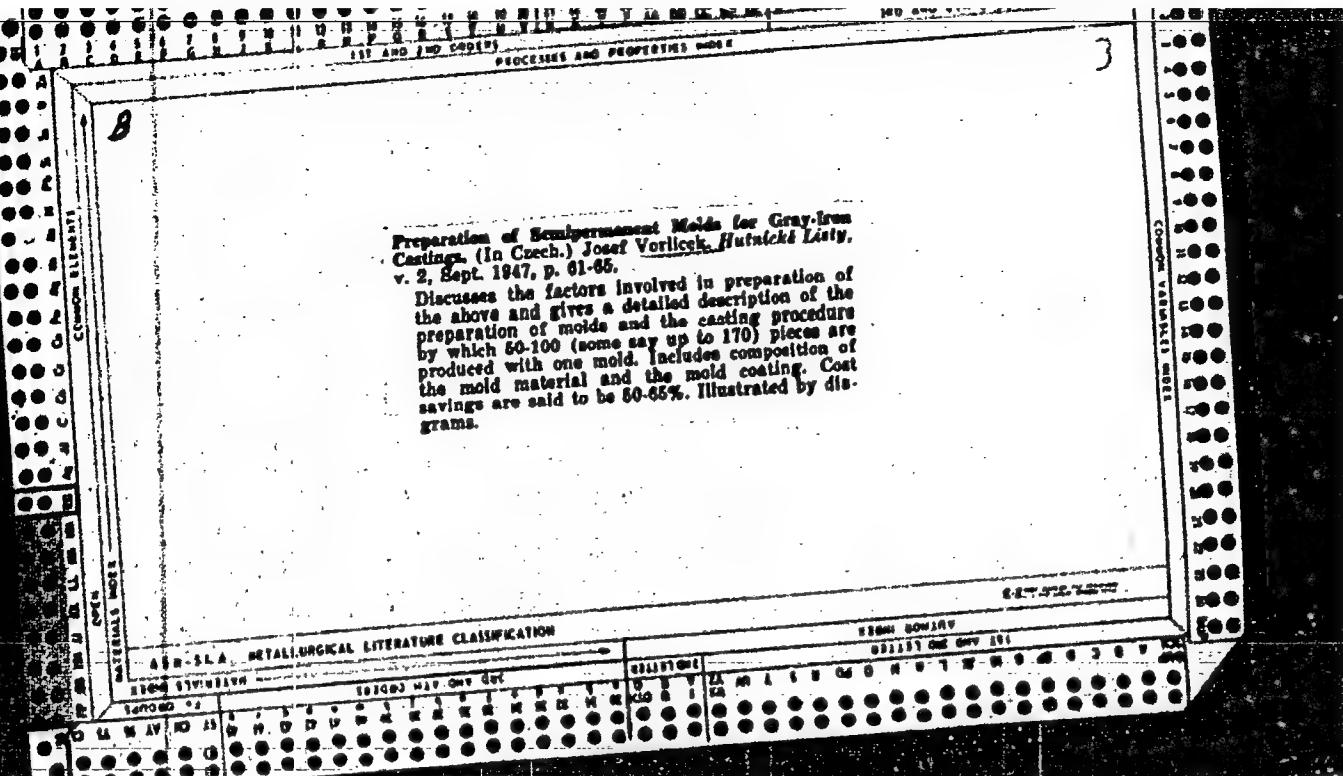
Vorlincek, J., Sekarka, I. Use of complexones in chemical analysis." XXXVII. Determination of uranium by the titration of ammonia with hypobromite. p. 512 CASOPIS PRO PESTOVANI MATEMATIKY. CZECHOSLOVAK MATHEMATICAL JOURNAL. Vol. 47, no. 4, Apr. 1953. Praha, Czechoslovakia.

SO: Monthly List of East European Accessions, LC., Vol. 3, No. 1, Jan. 1954, Uncl.

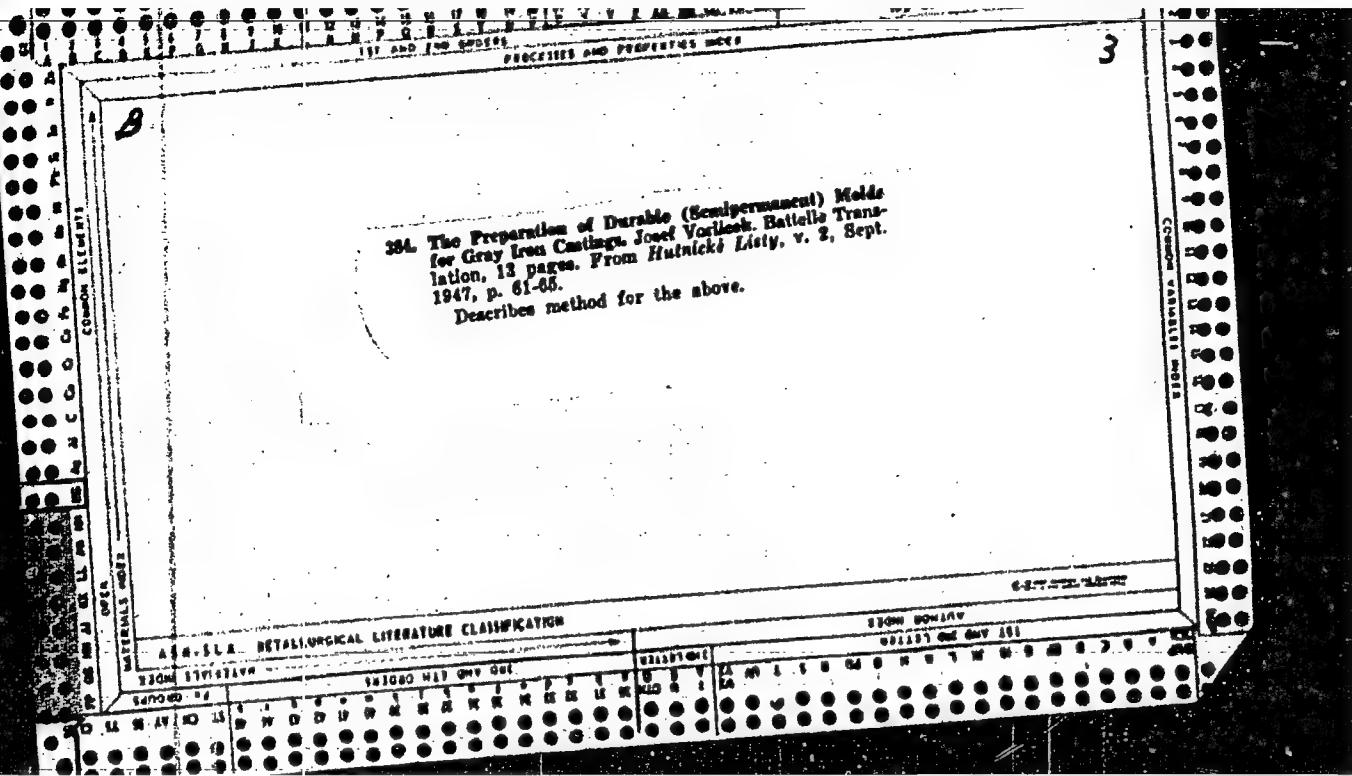


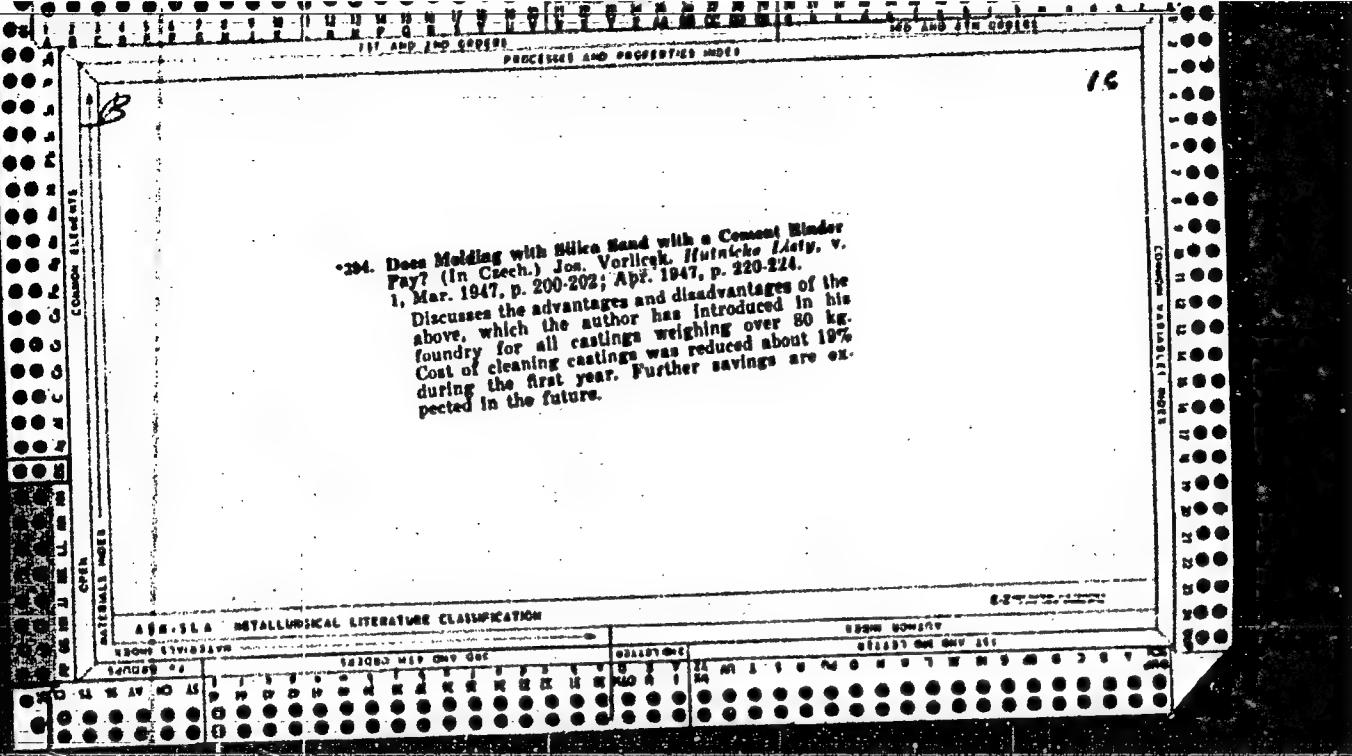
*Preparation of Semipermanent Molds for Gray-Iron Castings. (In Czech.) Josef Vorlický, Hutmácký Lisy, v. 2, Sept. 1847, p. 61-65.*

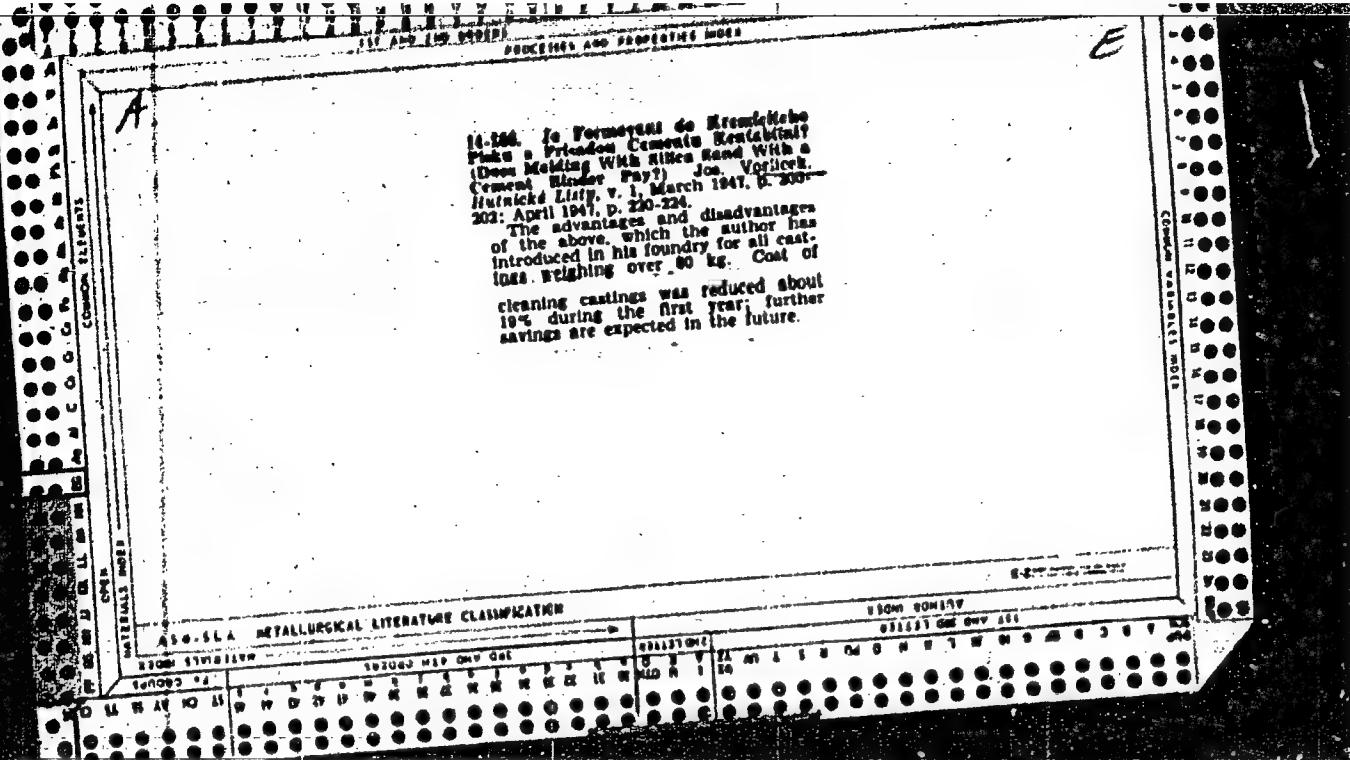
2, Sept. 1947, p. 61-66.  
Discusses the factors involved in preparation of the above and gives a detailed description of the preparation of molds and the casting procedure by which 50-100 (some say up to 170) pieces are produced with one mold. Includes composition of the mold material and the mold coating. Cost savings are said to be 50-65%. Illustrated by diagrams.



384. The Preparation of Durable (Semipermanent) Molds for Gray Iron Castings. Josef Vorlcek. Battelle Translation, 13 pages. From Hutnicki Listy, v. 2, Sept. 1947, p. 61-68.  
Describes method for the above.







S/263/62/000/018/003/006  
I031/I242

AUTHOR: Vorliček, Ivo

TITLE: Linear transistorized voltmeter

PERIODICAL: Referativnyy zhurnal, Otdelnyy vypusk. 32.  
Izmeritel'naya tekhnika, no. 18, 1962, 44,  
abstract 32.18.313. (Automatizace, v.5, no. 2,  
1962, 47 [Czech] )

TEXT: A small transistorized voltmeter has been developed by the Návika National Enterprises of Czechoslovakia. The voltmeter is very accurate, inexpensive, of simple design, and robust construction. The scale is uniform, the error not exceeding  $\pm 1\%$ . Permissible fluctuation of the supply voltage is  $\pm 10\%$ . The input resistance of the instrument is  $\sim 15$  kohm. A special feature

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S/263/62/000/018/003/006  
I031/I242

Linear transistorized....

of the voltmeter is the absence of electrical zero adjuster. A full description of the circuit diagram is given, including data for all the elements except those in the supply transformer, along with turning and adjusting instructions. The instrument is designed for measuring voltages ranging from 10 to 100 mV, the current requirement is ~ 40 mA, at 220V ac. ✓

[Abstracter's note: Complete translation.]

Card 2/2

VORLICEK, Ivo

Transistorized linear A.C. voltmeter. Automatizace 5 no.2:47 P '62.

1. Navika, n.p., Praha.

VORLICEK, I.

Phasing four-terminal networks with constant amplitude transfer. p. 67.  
SLABOPROUDY OBZOR, Prague, Vol. 15, no. 2, Feb. 1954.

SO: Monthly List of East European Accession, (EEAL), LC, Vol. 5, no. 6 June 1956, Uncl.